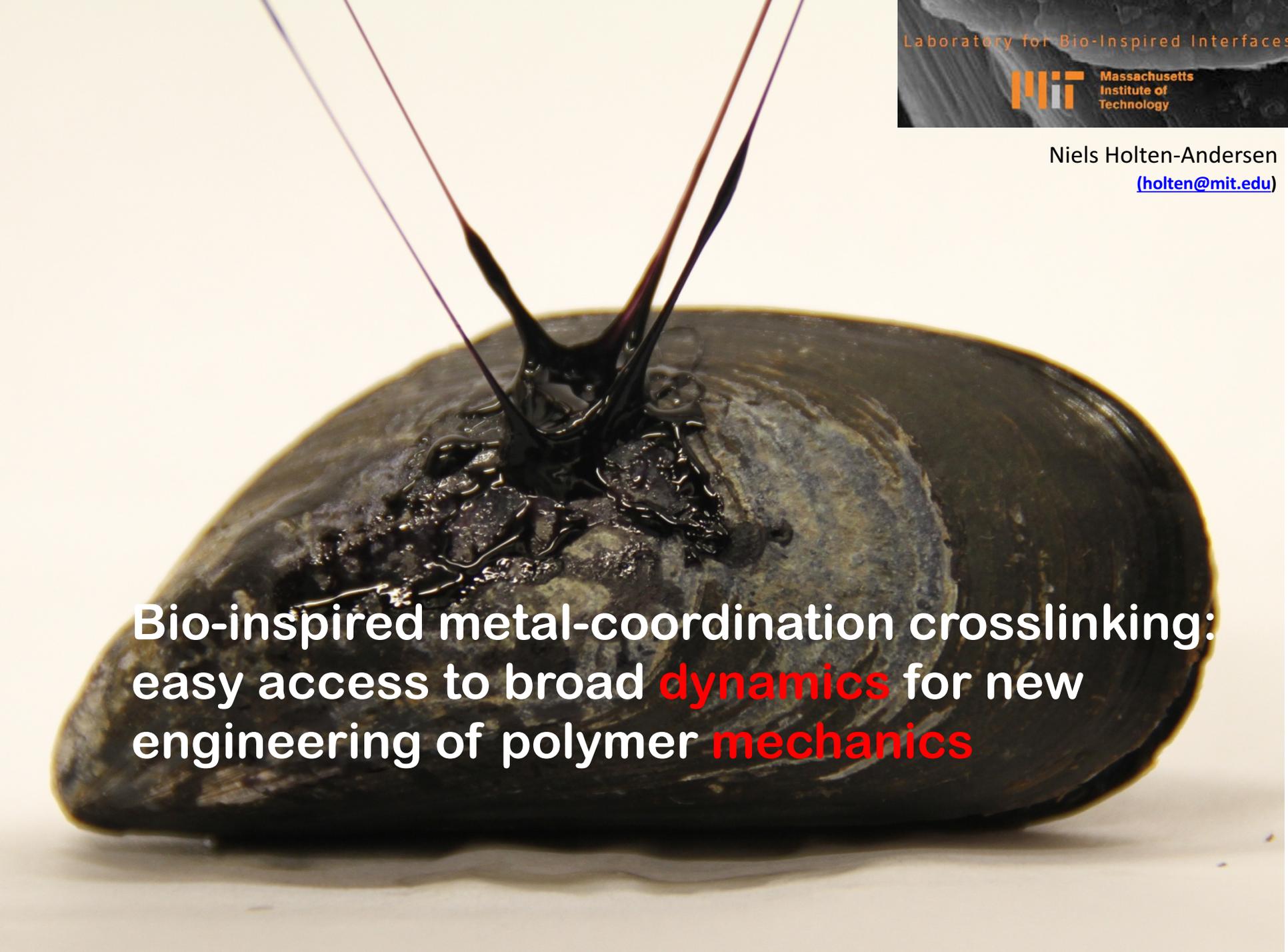
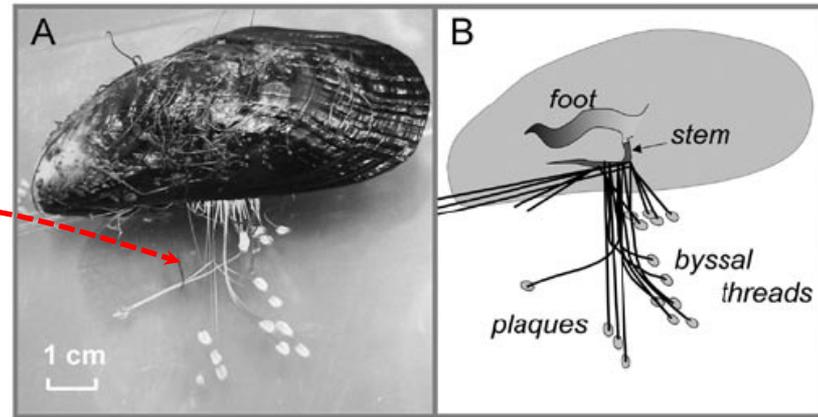


**Bio-inspired metal-coordination crosslinking:
easy access to broad dynamics for new
engineering of polymer mechanics**



Bio-inspired metal-coordination crosslinking:
easy access to broad **dynamics** for new
engineering of polymer **mechanics**

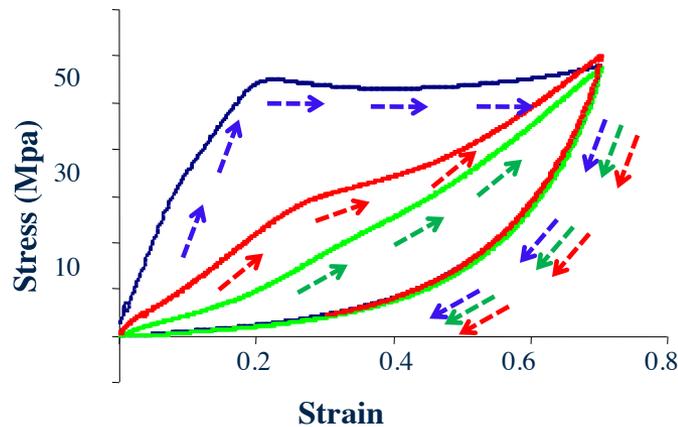
Mechanical energy dissipation in mussel threads



Holten-Andersen & Waite, *J. Dental Research*. **87**, 701-709, (2008)

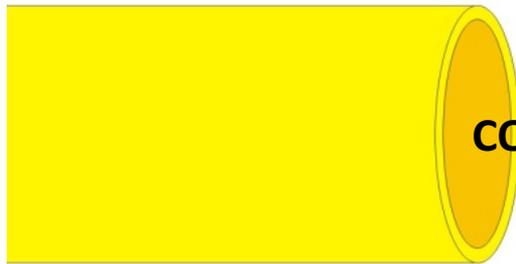


Repeated stress-strain cycles



Toughness
Self-healing

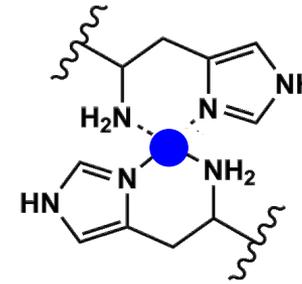
Metal-ion coordinate crosslinking in mussel threads



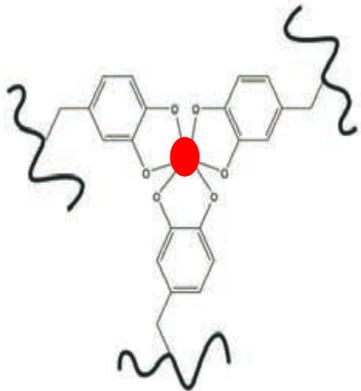
CORE

COATING

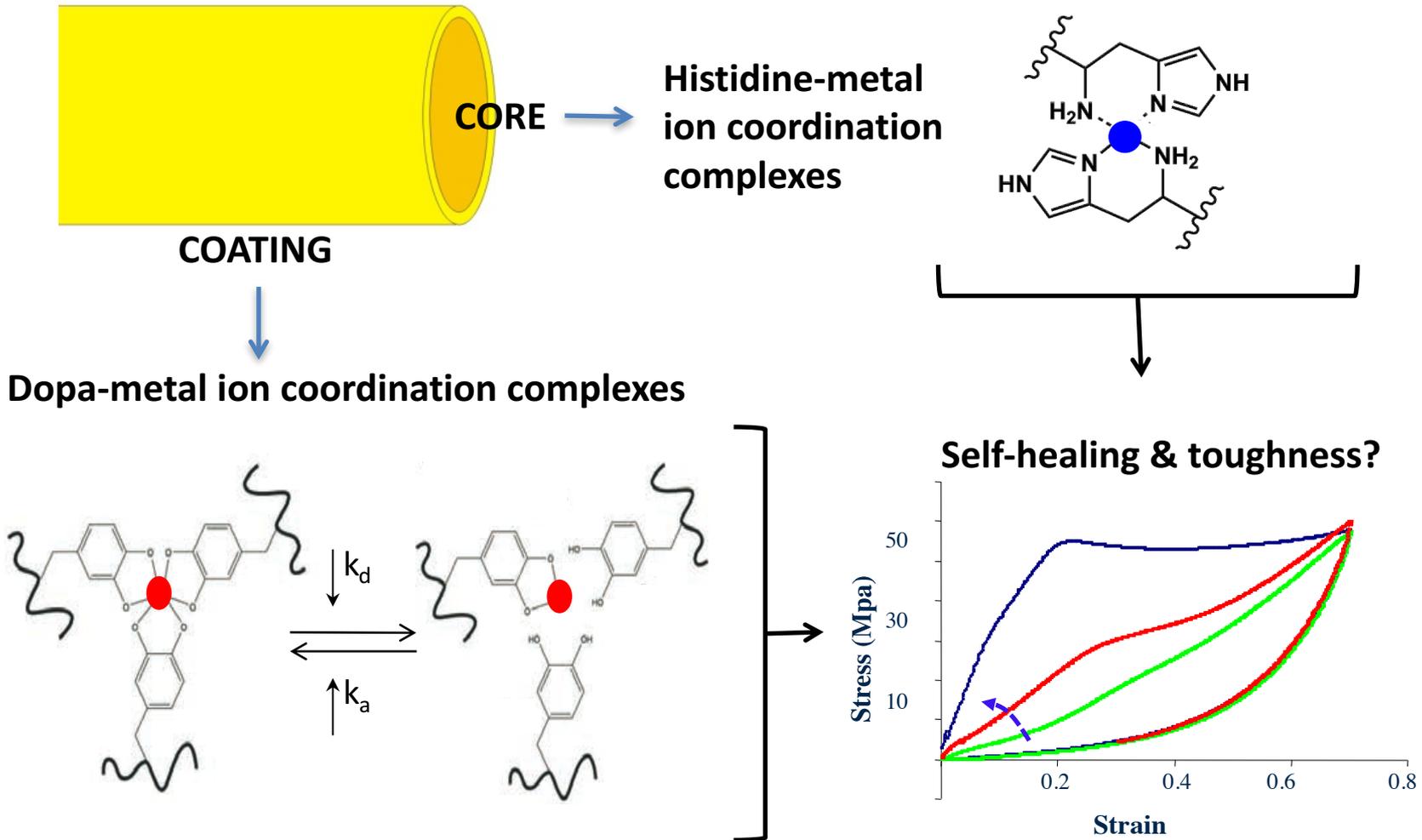
Histidine-metal
ion coordination
complexes



Dopa-metal ion coordination complexes

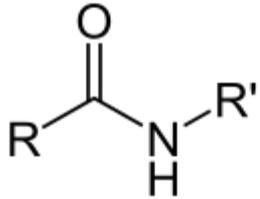


Metal-ion coordinate crosslinking in mussel threads



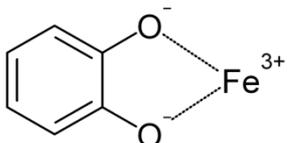
Metal-coordination in Nature

Covalent bonds



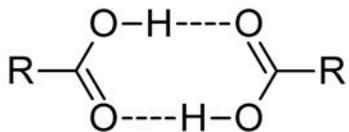
~100-1000 kJ/mol

Coordination bonds



~10-100 kJ/mol

Non-covalent bonds

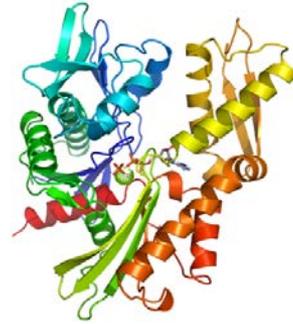


~1-10 kJ/mol

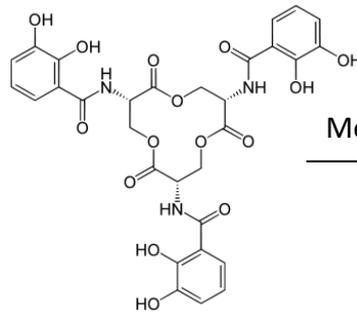
Metallo-proteins

Siderophores

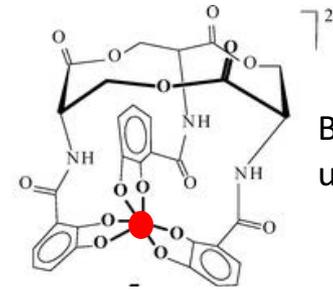
Bio-materials



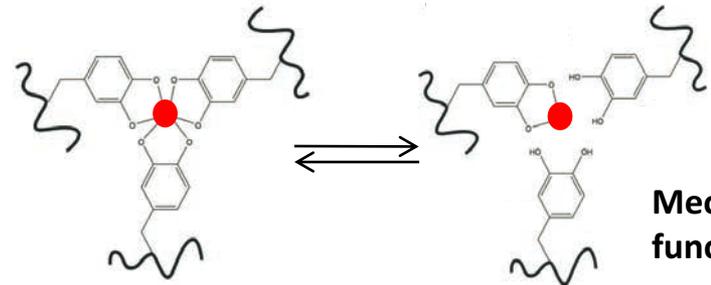
Enzymes
Fe-transport
O₂-transport
Heat shock proteins



Metal



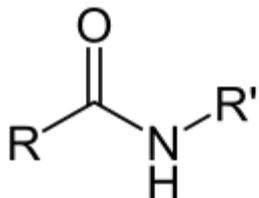
Bacterial metal-uptake



Mechanical function?

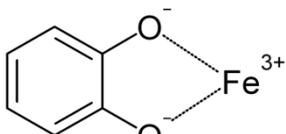
Metal-coordination in Nature

Covalent bonds



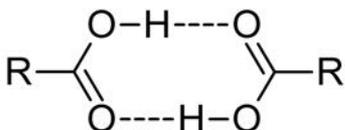
~100-1000 kJ/mol

Coordination bonds



~10-100 kJ/mol
(~4-40 kT/bond)

Non-covalent bonds

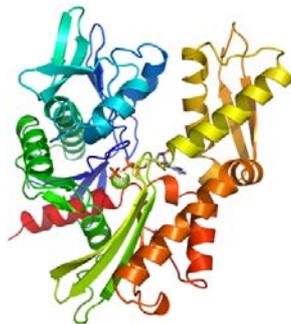


~1-10 kJ/mol

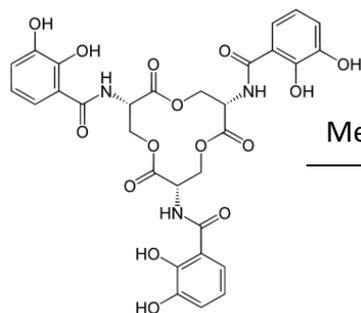
Metallo-proteins

Siderophores

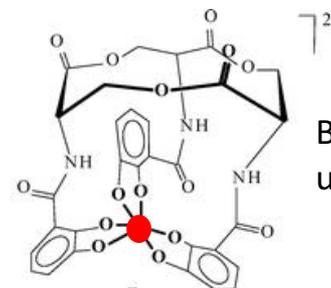
Bio-materials



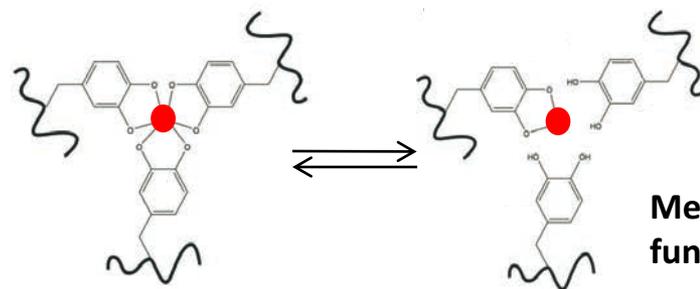
Enzymes
Fe-transport
O₂-transport
Heat shock proteins



Metal



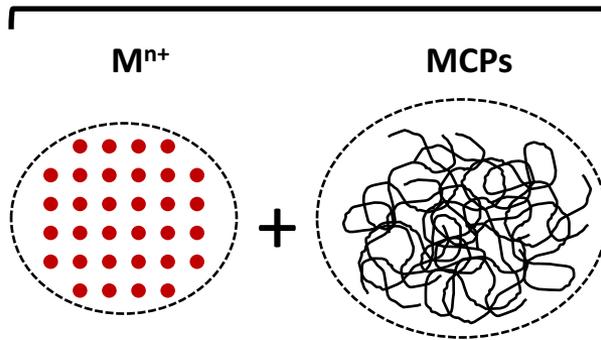
Bacterial metal-uptake



Mechanical function?

Metal-coordinated mussel thread assembly?

Intracellular storage at low pH (≤ 5)

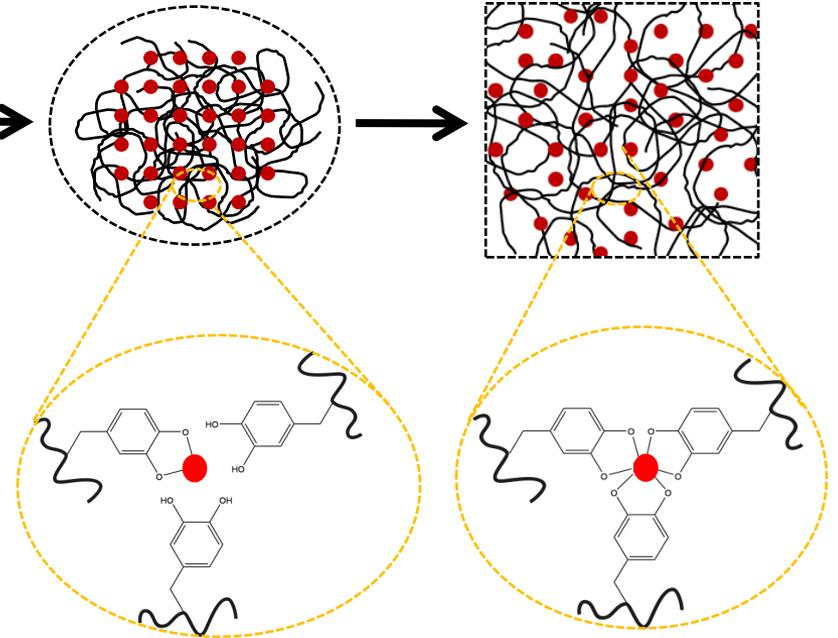
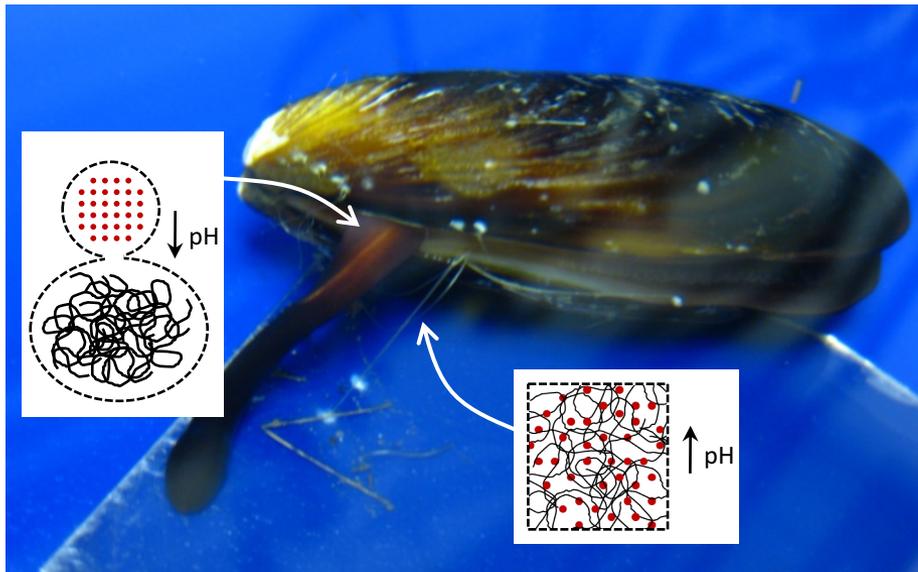


Mixing pH (≤ 5)

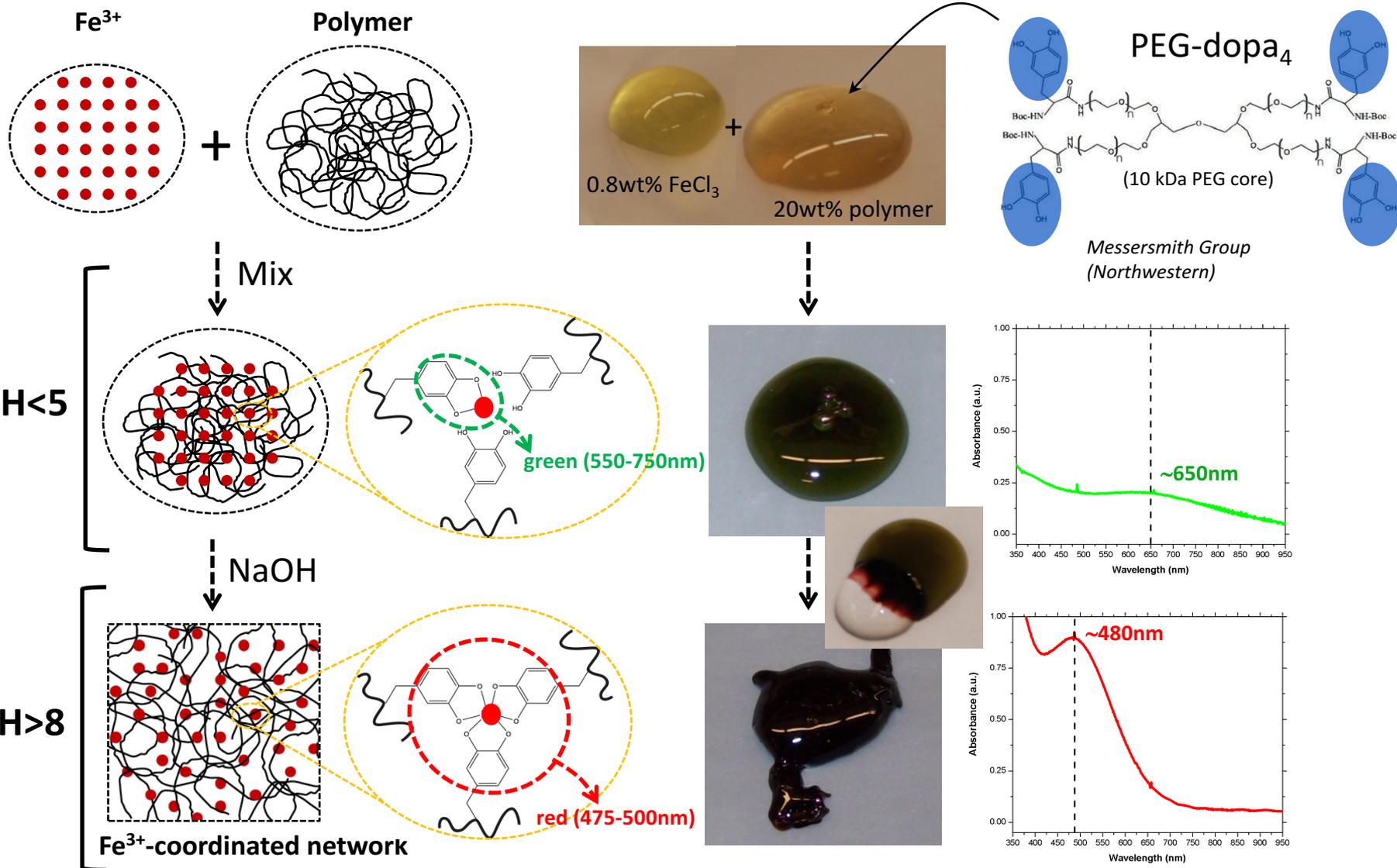
MCPs + M^{n+} mixture

Exposure to seawater (pH ≈ 8)

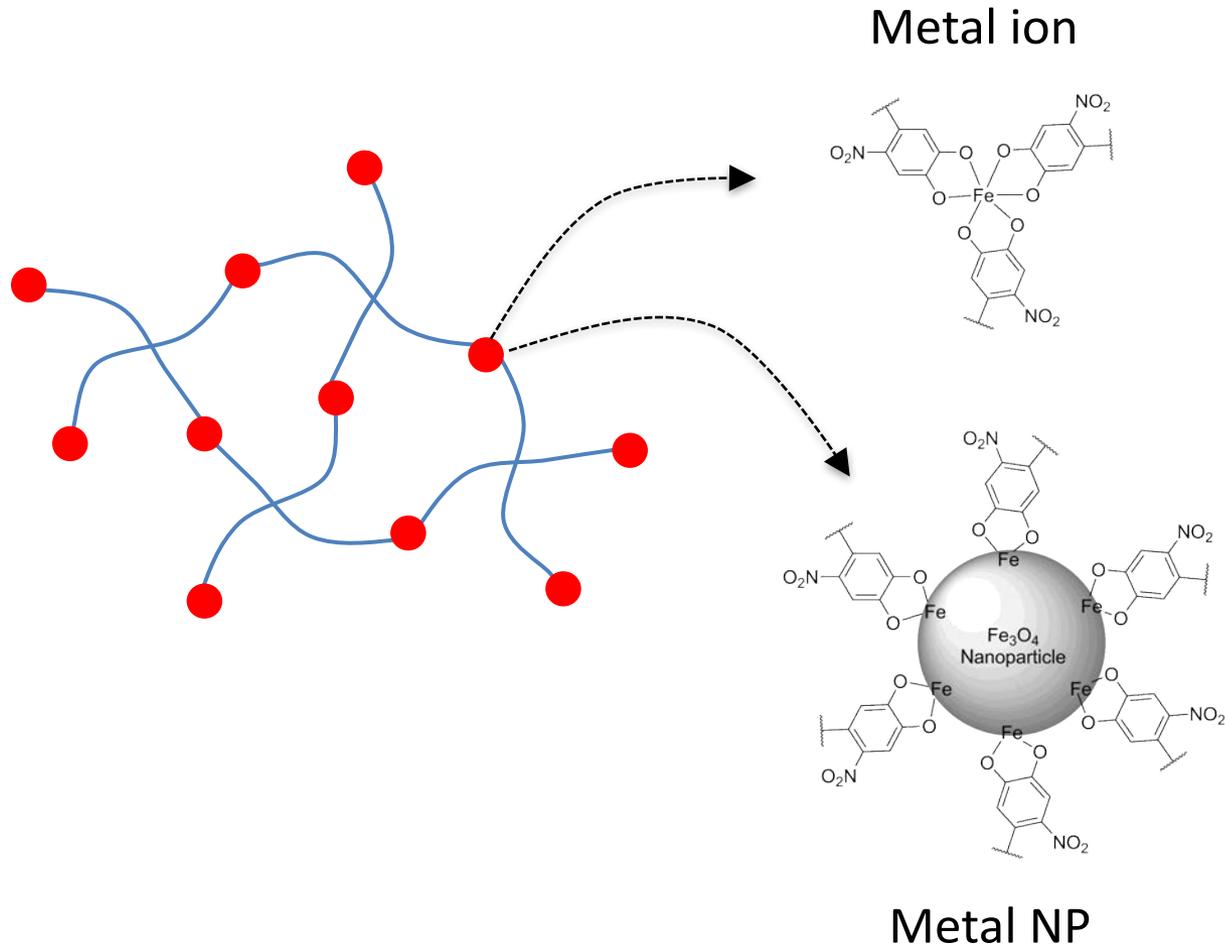
M^{n+} -coordinated network



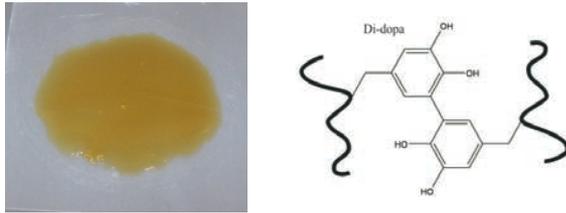
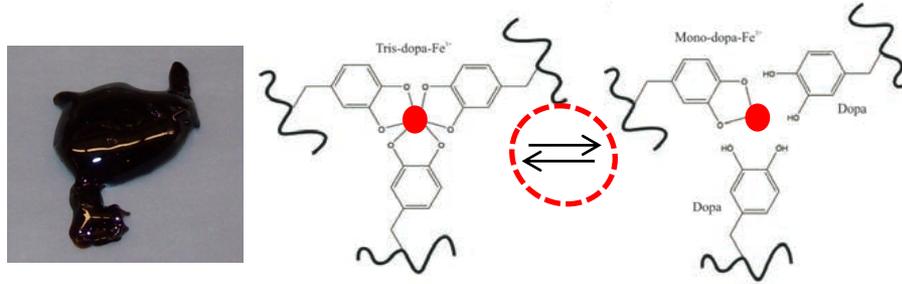
Metal-coordinated polymer network assembly



Bio-inspired metal-coordinate crosslinked networks



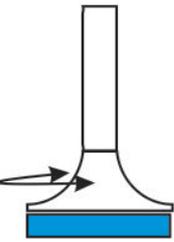
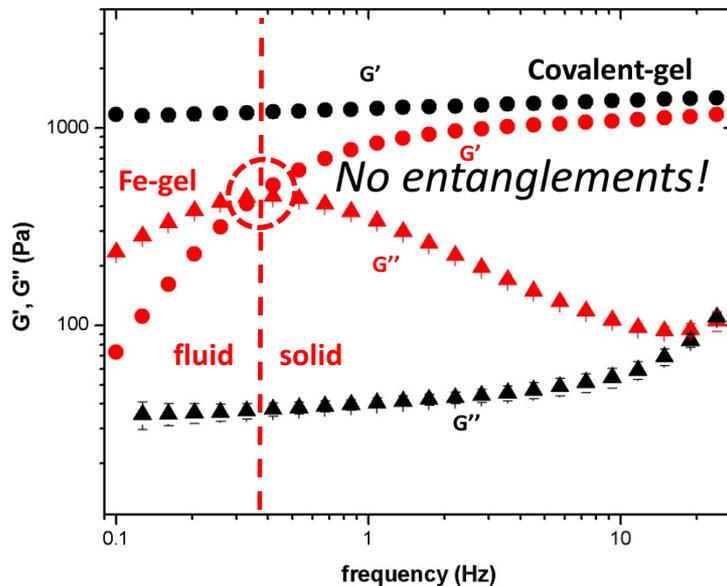
Metal ion-coordinate dynamics set gel mechanics



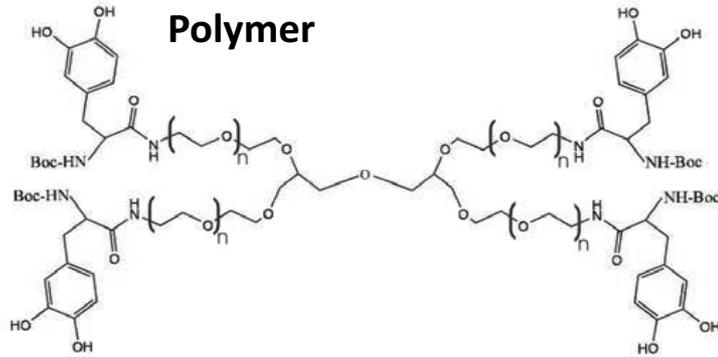
Elastic



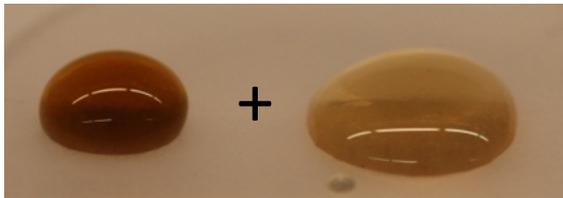
Visco-elastic



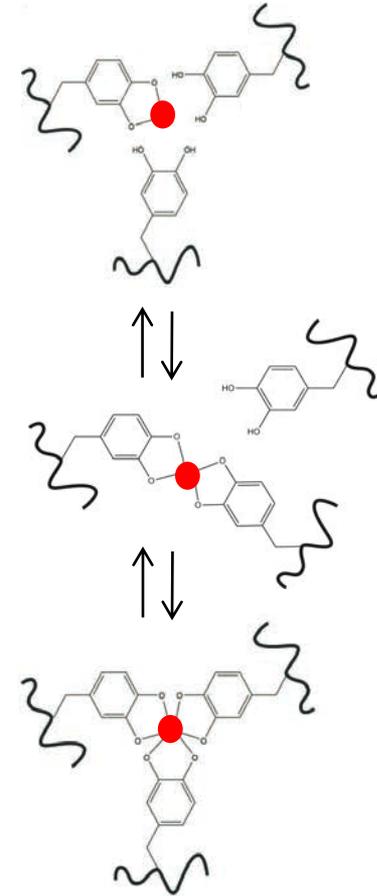
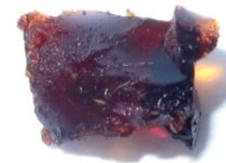
Metal ion-coordinate dynamics set gel mechanics



Metal



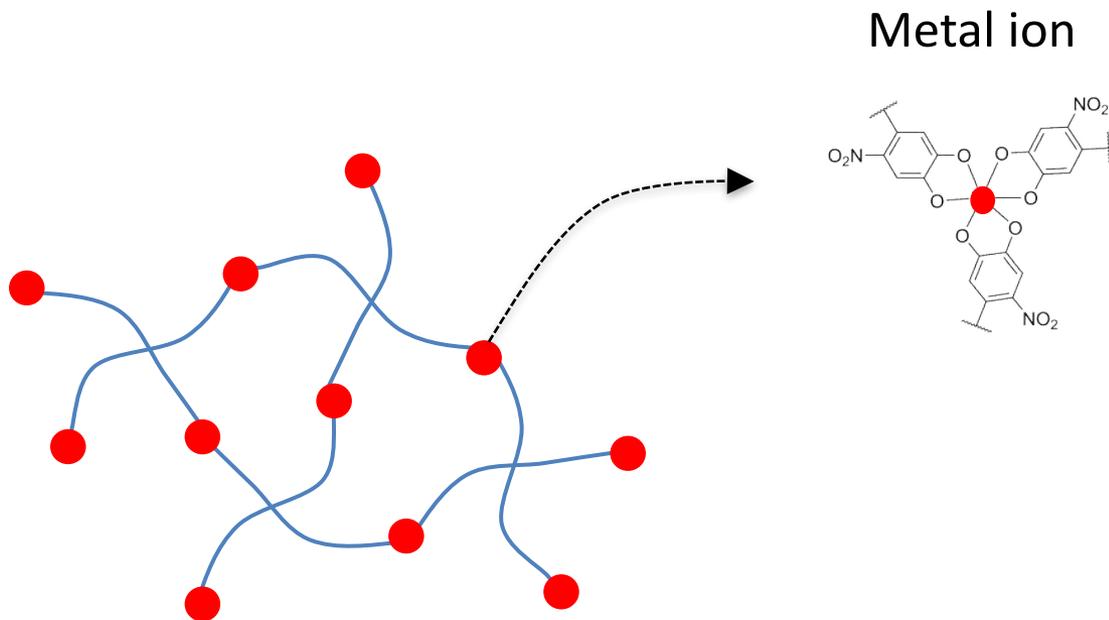
pH



Periodic Table of the Elements

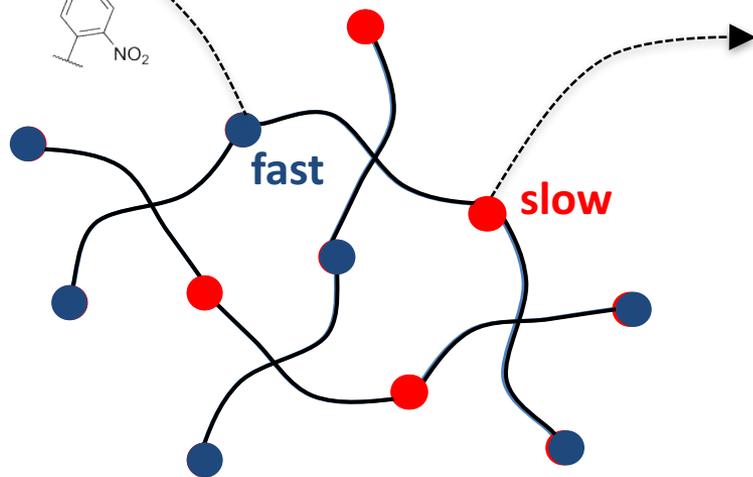
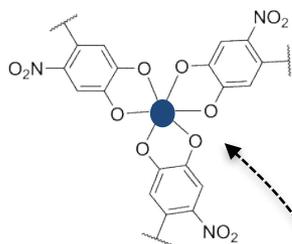
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| 1 IA H Hydrogen | 2 IIA He Helium | | | | | | | | | | | | | | | | | 18 VIIIA Ar Argon |
| 3 IIIA Li Lithium | 4 IVA Be Beryllium | | | | | | | | | | | 13 IIIA B Boron | 14 IVA C Carbon | 15 VA N Nitrogen | 16 VIA O Oxygen | 17 VIIA F Fluorine | 18 VIIIA Ne Neon | |
| 11 IA Na Sodium | 12 IIA Mg Magnesium | | | | | | | | | | | 31 IIIA Al Aluminum | 32 IVA Si Silicon | 33 VA P Phosphorus | 34 VIA S Sulfur | 35 VIIA Cl Chlorine | 36 VIIIA Ar Argon | |
| 19 IA K Potassium | 20 IIA Ca Calcium | | | | | | | | | | | 47 IIIA Ga Gallium | 48 IVA Ge Germanium | 49 VA As Arsenic | 50 VIA Se Selenium | 51 VIIA Br Bromine | 52 VIIIA Kr Krypton | |
| 37 IA Rb Rubidium | 38 IIA Sr Strontium | | | | | | | | | | | 63 IIIA In Indium | 64 IVA Sn Tin | 65 VA Sb Antimony | 66 VIA Te Tellurium | 67 VIIA I Iodine | 68 VIIIA Xe Xenon | |
| 55 IA Cs Cesium | 56 IIA Ba Barium | | | | | | | | | | | 81 IIIA Tl Thallium | 82 IVA Pb Lead | 83 VA Bi Bismuth | 84 VIA Po Polonium | 85 VIIA At Astatine | 86 VIIIA Rn Radon | |
| 87 IA Fr Francium | 88 IIA Ra Radium | | | | | | | | | | | 113 IIIA Nh Nihonium | 114 IVA Fl Flerovium | 115 VA Uup Ununpentium | 116 VIA Lv Livermorium | 117 VIIA Uus Ununseptium | 118 VIIIA Uuo Ununoctium | |
| <p>Lanthanide Series</p> <p>57 La Lanthanum, 58 Ce Cerium, 59 Pr Praseodymium, 60 Nd Neodymium, 61 Pm Promethium, 62 Sm Samarium, 63 Eu Europium, 64 Gd Gadolinium, 65 Tb Terbium, 66 Dy Dysprosium, 67 Ho Holmium, 68 Er Erbium, 69 Tm Thulium, 70 Yb Ytterbium, 71 Lu Lutetium</p> <p>Actinide Series</p> <p>89 Ac Actinium, 90 Th Thorium, 91 Pa Protactinium, 92 U Uranium, 93 Np Neptunium, 94 Pu Plutonium, 95 Am Americium, 96 Cm Curium, 97 Bk Berkelium, 98 Cf Californium, 99 Es Einsteinium, 100 Fm Fermium, 101 Md Mendelevium, 102 No Nobeium, 103 Lr Lawrencium</p> <p>Alkali Metal, Alkaline Earth, Transition Metal, Basic Metal, Semimetal, Nonmetal, Halogen, Noble Gas, Lanthanide, Actinide</p> | | | | | | | | | | | | | | | | | | |

Metal ion-coordinate dynamics set gel mechanics

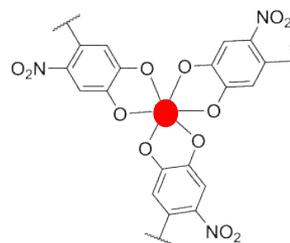


Metal ion-coordinate dynamics set gel mechanics

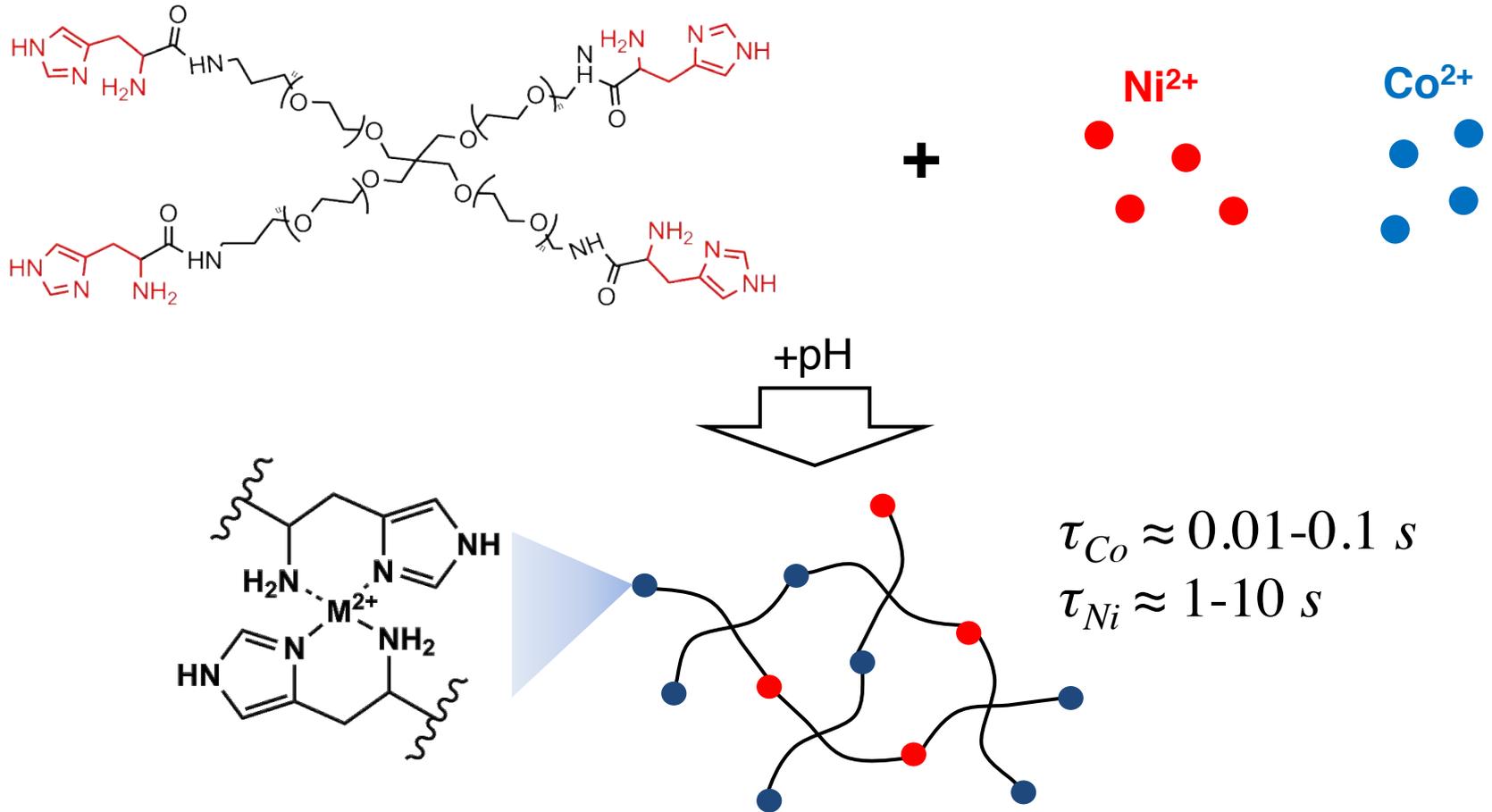
Metal ion B



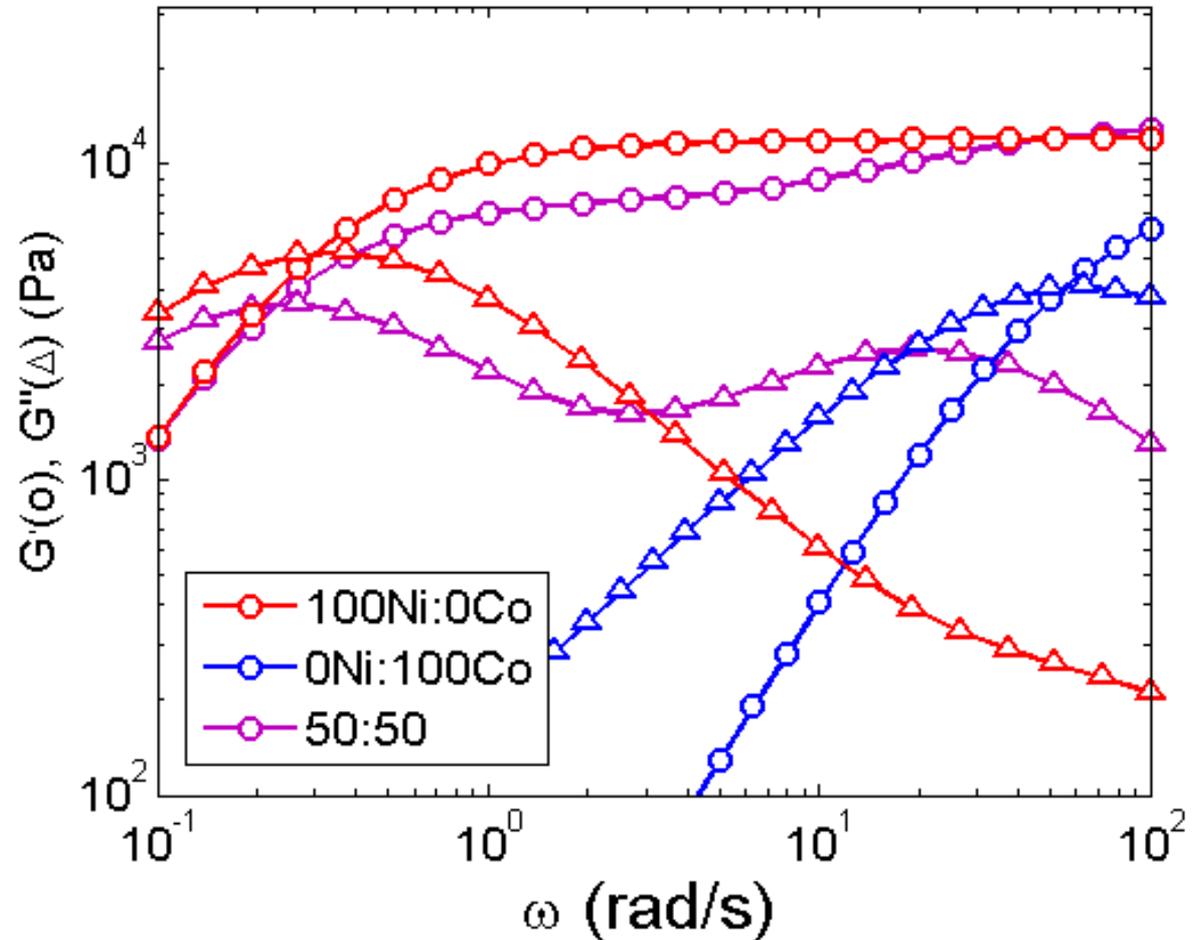
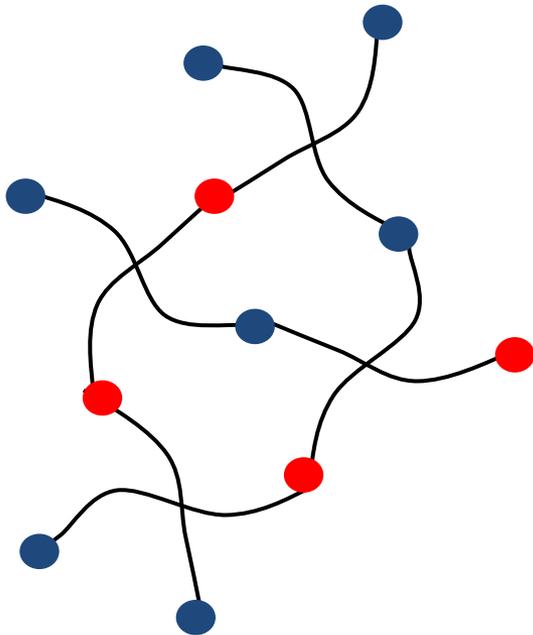
Metal ion A



Metal ion-coordinate dynamics set gel mechanics

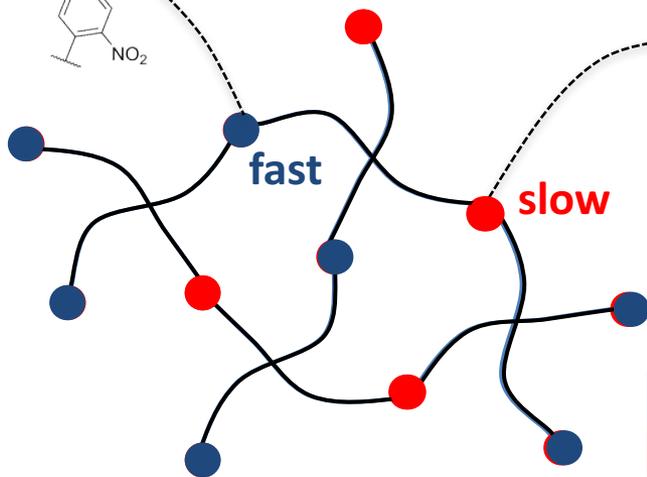
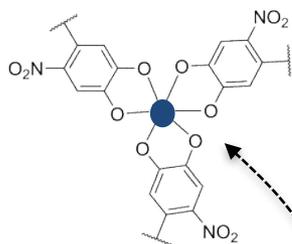


Metal ion-coordinate dynamics set gel mechanics

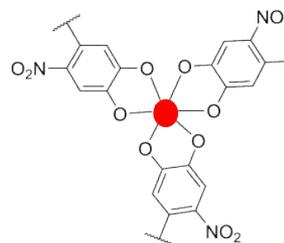


Metal ion-coordinate dynamics set gel mechanics

Metal ion B



Metal ion A

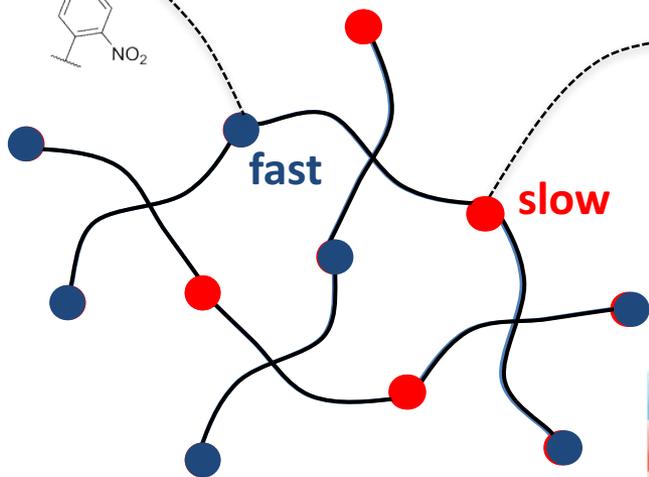
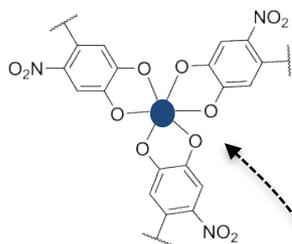


Periodic Table of the Elements

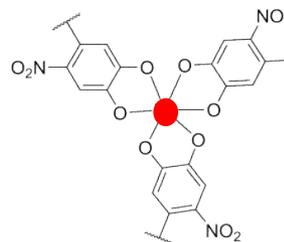
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| 1 IA H Hydrogen 1.008 | 2 IIA Be Beryllium 9.012 | | | | | | | | | | | 13 IIIA Al Aluminum 26.982 | 14 IVA Si Silicon 28.086 | 15 VA P Phosphorus 30.974 | 16 VIA S Sulfur 32.06 | 17 VIIA Cl Chlorine 35.45 | 18 VIIIA Ar Argon 39.948 | |
| 3 IB Li Lithium 6.941 | 4 IIA Be Beryllium 9.012 | 5 IIIB Sc Scandium 44.956 | 6 IVB Ti Titanium 47.88 | 7 VB V Vanadium 50.942 | 8 VIB Cr Chromium 51.996 | 9 VIIB Mn Manganese 54.938 | 10 VIII Fe Iron 55.845 | 11 VIII Co Cobalt 58.933 | 12 VIII Ni Nickel 58.693 | 13 IB Cu Copper 63.546 | 14 IIB Zn Zinc 65.38 | 15 IIIB Ga Gallium 69.723 | 16 IVB Ge Germanium 72.63 | 17 VB As Arsenic 74.922 | 18 VIB Se Selenium 78.96 | 19 VIIB Br Bromine 79.904 | 20 VIIIB Kr Krypton 83.798 | |
| 19 IIA K Potassium 39.098 | 20 IIA Ca Calcium 40.078 | 21 IIIB Sc Scandium 44.956 | 22 IVB Ti Titanium 47.88 | 23 VB V Vanadium 50.942 | 24 VIB Cr Chromium 51.996 | 25 VIIB Mn Manganese 54.938 | 26 VIII Fe Iron 55.845 | 27 VIII Co Cobalt 58.933 | 28 VIII Ni Nickel 58.693 | 29 IB Cu Copper 63.546 | 30 IIB Zn Zinc 65.38 | 31 IIIB Ga Gallium 69.723 | 32 IVB Ge Germanium 72.63 | 33 VB As Arsenic 74.922 | 34 VIB Se Selenium 78.96 | 35 VIIB Br Bromine 79.904 | 36 VIIIB Kr Krypton 83.798 | |
| 37 IIA Rb Rubidium 85.468 | 38 IIA Sr Strontium 87.62 | 39 IIIB Y Yttrium 88.906 | 40 IVB Zr Zirconium 91.224 | 41 VB Nb Niobium 92.906 | 42 VIB Mo Molybdenum 95.94 | 43 VIIB Tc Technetium 98.906 | 44 VIII Ru Ruthenium 101.07 | 45 VIII Rh Rhodium 102.905 | 46 VIII Pd Palladium 106.42 | 47 IB Ag Silver 107.868 | 48 IIB Cd Cadmium 112.411 | 49 IIIB In Indium 114.818 | 50 IVB Sn Tin 118.710 | 51 VB Sb Antimony 121.757 | 52 VIB Te Tellurium 127.6 | 53 VIIB I Iodine 126.905 | 54 VIIIB Xe Xenon 131.29 | |
| 55 IIA Cs Cesium 132.905 | 56 IIA Ba Barium 137.327 | 57-71 Lanthanide Series | 72 IVB Hf Hafnium 178.49 | 73 VB Ta Tantalum 180.948 | 74 VIB W Tungsten 183.84 | 75 VIIB Re Rhenium 186.207 | 76 VIII Os Osmium 190.23 | 77 VIII Ir Iridium 192.222 | 78 VIII Pt Platinum 195.084 | 79 IB Au Gold 196.967 | 80 IIB Hg Mercury 200.59 | 81 IIIB Tl Thallium 204.38 | 82 IVB Pb Lead 207.2 | 83 VB Bi Bismuth 208.98 | 84 VIB Po Polonium 209 | 85 VIIB At Astatine 209 | 86 VIIIB Rn Radon 222 | |
| 87 IIA Fr Francium 223 | 88 IIA Ra Radium 226 | 89-103 Actinide Series | 104 Rf Rutherfordium 261 | 105 Db Dubnium 262 | 106 Sg Seaborgium 263 | 107 Bh Bohrium 264 | 108 Hs Hassium 265 | 109 Mt Meitnerium 266 | 110 Ds Darmstadtium 267 | 111 Rg Roentgenium 268 | 112 Cn Copernicium 269 | 113 Uut Ununtrium 270 | 114 Fl Flerovium 271 | 115 Uup Ununpentium 272 | 116 Lv Livermorium 273 | 117 Uus Ununseptium 274 | 118 Uuo Ununoctium 276 | |
| Lanthanide Series | | 57 La Lanthanum 138.905 | 58 Ce Cerium 140.12 | 59 Pr Praseodymium 140.908 | 60 Nd Neodymium 144.24 | 61 Pm Promethium 144.913 | 62 Sm Samarium 150.36 | 63 Eu Europium 151.964 | 64 Gd Gadolinium 157.25 | 65 Tb Terbium 158.925 | 66 Dy Dysprosium 162.50 | 67 Ho Holmium 164.930 | 68 Er Erbium 167.259 | 69 Tm Thulium 168.930 | 70 Yb Ytterbium 173.054 | 71 Lu Lutetium 174.967 | | |
| Actinide Series | | 89 Ac Actinium 227 | 90 Th Thorium 232.038 | 91 Pa Protactinium 231.036 | 92 U Uranium 238.029 | 93 Np Neptunium 237.048 | 94 Pu Plutonium 244.064 | 95 Am Americium 243.061 | 96 Cm Curium 247.070 | 97 Bk Berkelium 247.070 | 98 Cf Californium 251.083 | 99 Es Einsteinium 252.083 | 100 Fm Fermium 257.105 | 101 Md Mendelevium 258.10 | 102 No Nobelium 259.10 | 103 Lr Lawrencium 260.10 | | |
| Alkali Metal | | Alkaline Earth | | Transition Metal | | | Block | | Semimetal | | Nonmetal | | Halogen | | Noble Gas | | Actinide | |

Metal ion-coordinate dynamics set gel mechanics

Lanthanide Metal ion B



Lanthanide metal ion A

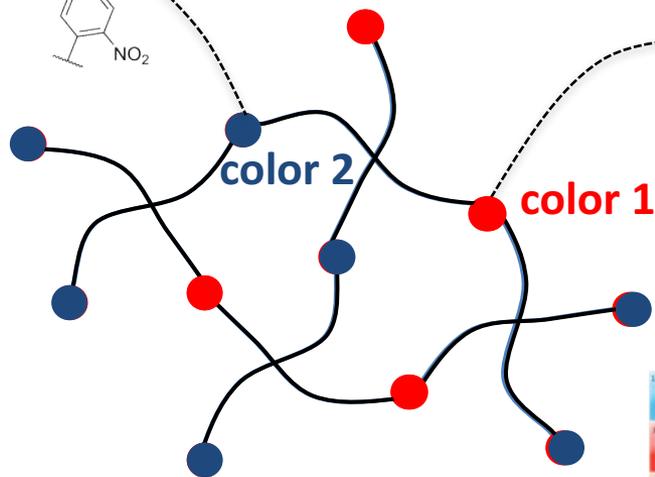
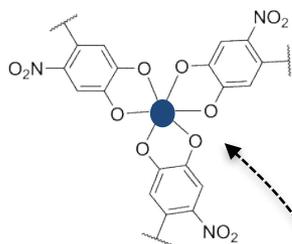


Periodic Table of the Elements

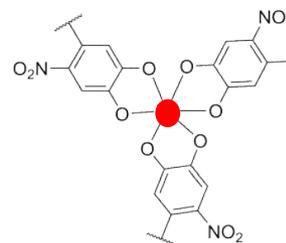
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|--------------------------------------|---------------------------------------|---|--|---------------------------------------|--|--|---|--|---|-------------------------------------|---------------------------------------|---|---|---------------------------------------|---------------------------------------|---------------------------------------|--|------------------|-------|-----------|----------|---------|-----------|----------|
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| 19 IA K Potassium 39.098 | 20 IIA Ca Calcium 40.078 | 21 IIIB Sc Scandium 44.956 | 22 IVB Ti Titanium 47.88 | 23 VB V Vanadium 50.942 | 24 VIB Cr Chromium 51.996 | 25 VIIB Mn Manganese 54.938 | 26 VIII Fe Iron 55.845 | 27 VIII Co Cobalt 58.933 | 28 VIII Ni Nickel 58.69 | 29 IB Cu Copper 63.546 | 30 IIB Zn Zinc 65.38 | 31 IIIB Ga Gallium 69.723 | 32 IIIVB Ge Germanium 72.63 | 33 VB As Arsenic 74.922 | 34 VIB Se Selenium 78.96 | 35 VIIB Br Bromine 79.904 | 36 VIIIB Kr Krypton 83.798 | | | | | | | |
| 37 IA Rb Rubidium 85.468 | 38 IIA Sr Strontium 87.62 | 39 IIIB Y Yttrium 88.906 | 40 IVB Zr Zirconium 91.224 | 41 VB Nb Niobium 92.906 | 42 VIB Mo Molybdenum 95.94 | 43 VIIB Tc Technetium 98.906 | 44 VIII Ru Ruthenium 101.07 | 45 VIII Rh Rhodium 102.905 | 46 VIII Pd Palladium 106.42 | 47 IB Ag Silver 107.868 | 48 IIB Cd Cadmium 112.411 | 49 IIIB In Indium 114.818 | 50 IIIVB Sn Tin 118.710 | 51 VB Sb Antimony 121.757 | 52 VIB Te Tellurium 127.6 | 53 VIIB I Iodine 126.905 | 54 VIIIB Xe Xenon 131.29 | | | | | | | |
| 55 IA Cs Cesium 132.905 | 56 IIA Ba Barium 137.327 | 57-71 Lanthanide Series La Lanthanum 138.905 | 72 IVB Hf Hafnium 178.49 | 73 VB Ta Tantalum 180.948 | 74 VIB W Tungsten 183.84 | 75 VIIB Re Rhenium 186.207 | 76 VIII Os Osmium 190.23 | 77 VIII Ir Iridium 192.222 | 78 VIII Pt Platinum 195.084 | 79 IB Au Gold 196.967 | 80 IIB Hg Mercury 200.59 | 81 IIIB Tl Thallium 204.384 | 82 IIIVB Pb Lead 207.2 | 83 VB Bi Bismuth 208.980 | 84 VIB Po Polonium 209 | 85 VIIB At Astatine 210 | 86 VIIIB Rn Radon 222 | | | | | | | |
| 87 IA Fr Francium 223 | 88 IIA Ra Radium 226 | 89-103 Lanthanide Series La Lanthanum 138.905 | 104 Rf Rutherfordium 261 | 105 Db Dubnium 262 | 106 Sg Seaborgium 263 | 107 Bh Bohrium 264 | 108 Hs Hassium 265 | 109 Mt Meitnerium 266 | 110 Ds Darmstadtium 267 | 111 Dsb Roentgenium 268 | 112 Cn Copernicium 269 | 113 Uut Utenhium 270 | 114 Fl Flerovium 271 | 115 Uup Ununpentium 272 | 116 Lv Livermorium 273 | 117 Uus Ununseptium 274 | 118 Uuo Ununoctium 276 | | | | | | | |
| Lanthanide Series | | 57 La Lanthanum 138.905 | 58 Ce Cerium 140.12 | 59 Pr Praseodymium 140.908 | 60 Nd Neodymium 144.24 | 61 Pm Promethium 144.913 | 62 Sm Samarium 150.36 | 63 Eu Europium 151.964 | 64 Gd Gadolinium 157.25 | 65 Tb Terbium 158.925 | 66 Dy Dysprosium 162.50 | 67 Ho Holmium 164.930 | 68 Er Erbium 167.259 | 69 Tm Thulium 168.934 | 70 Yb Ytterbium 173.054 | 71 Lu Lutetium 174.967 | Actinide Series | | | | | | | |
| 89 Ac Actinium 227 | 90 Th Thorium 232.038 | 91 Pa Protactinium 231.036 | 92 U Uranium 238.029 | 93 Np Neptunium 237.048 | 94 Pu Plutonium 244.064 | 95 Am Americium 243.061 | 96 Cm Curium 247.077 | 97 Bk Berkelium 247.077 | 98 Cf Californium 251.083 | 99 Es Einsteinium 252.083 | 100 Fm Fermium 257.103 | 101 Md Mendelevium 258.103 | 102 No Nobelium 259.103 | 103 Lr Lawrencium 260.103 | Alkali Metal | | Alkaline Earth | Transition Metal | Block | Semimetal | Nonmetal | Halogen | Noble Gas | Actinide |

Metal ion-coordinate dynamics set gel mechanics

Lanthanide Metal ion B



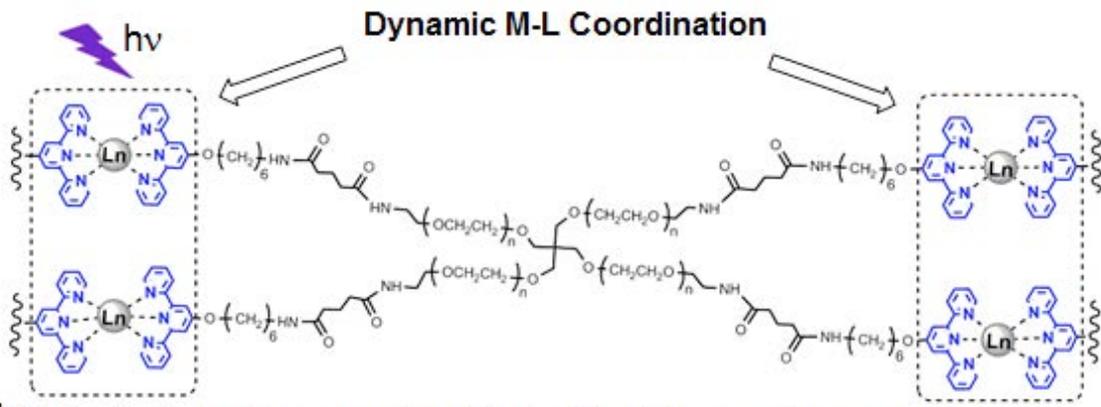
Lanthanide metal ion A



Periodic Table of the Elements

| | | | | | | | | | | | | | | | | | | | |
|-----------------------------------|--------------------------------------|----------------------------------|-----------------------------------|-------------------------------------|---------------------------------|-----------------------------------|----------------------------------|----------------------------------|----------------------------------|----------------------------------|------------------------------------|--|--------------------------------------|---------------------------------------|-----------------------------------|--|--------------------------------------|----------|--|
| 1 IA H Hydrogen 1.008 | 2 IIA Be Beryllium 9.012 | | | | | | | | | | | 13 IIIA Al Aluminum 26.982 | 14 IVA Si Silicon 28.086 | 15 VA P Phosphorus 30.974 | 16 VIA S Sulfur 32.06 | 17 VIIA Cl Chlorine 35.453 | 18 VIIIA Ar Argon 39.948 | | |
| 3 Li Lithium 6.941 | 4 Be Beryllium 9.012 | 5 B Boron 10.811 | 6 C Carbon 12.011 | 7 N Nitrogen 14.007 | 8 O Oxygen 15.999 | 9 F Fluorine 18.998 | 10 Ne Neon 20.180 | 11 Na Sodium 22.990 | 12 Mg Magnesium 24.305 | 13 Al Aluminum 26.982 | 14 Si Silicon 28.086 | 15 P Phosphorus 30.974 | 16 S Sulfur 32.06 | 17 Cl Chlorine 35.453 | 18 Ar Argon 39.948 | | | | |
| 19 K Potassium 39.098 | 20 Ca Calcium 40.078 | 21 Sc Scandium 44.956 | 22 Ti Titanium 47.88 | 23 V Vanadium 50.942 | 24 Cr Chromium 51.996 | 25 Mn Manganese 54.938 | 26 Fe Iron 55.845 | 27 Co Cobalt 58.933 | 28 Ni Nickel 58.693 | 29 Cu Copper 63.546 | 30 Zn Zinc 65.38 | 31 Ga Gallium 69.723 | 32 Ge Germanium 72.63 | 33 As Arsenic 74.922 | 34 Se Selenium 78.96 | 35 Br Bromine 79.904 | 36 Kr Krypton 83.80 | | |
| 37 Rb Rubidium 85.468 | 38 Sr Strontium 87.62 | 39 Y Yttrium 88.906 | 40 Zr Zirconium 91.224 | 41 Nb Niobium 92.906 | 42 Mo Molybdenum 95.94 | 43 Tc Technetium 98.906 | 44 Ru Ruthenium 101.07 | 45 Rh Rhodium 102.905 | 46 Pd Palladium 106.42 | 47 Ag Silver 107.868 | 48 Cd Cadmium 112.411 | 49 In Indium 114.818 | 50 Sn Tin 118.710 | 51 Sb Antimony 121.757 | 52 Te Tellurium 127.6 | 53 I Iodine 126.905 | 54 Xe Xenon 131.29 | | |
| 55 Cs Cesium 132.905 | 56 Ba Barium 137.327 | 57-71 Lanthanide Series | 72 Hf Hafnium 178.49 | 73 Ta Tantalum 180.948 | 74 W Tungsten 183.84 | 75 Re Rhenium 186.207 | 76 Os Osmium 190.23 | 77 Ir Iridium 192.222 | 78 Pt Platinum 195.084 | 79 Au Gold 196.967 | 80 Hg Mercury 200.59 | 81 Tl Thallium 204.38 | 82 Pb Lead 207.2 | 83 Bi Bismuth 208.98 | 84 Po Polonium 209 | 85 At Astatine 210 | 86 Rn Radon 222 | | |
| 87 Fr Francium 223 | 88 Ra Radium 226 | 89-103 Actinide Series | 104 Rf Rutherfordium 261 | 105 Db Dubnium 262 | 106 Sg Seaborgium 263 | 107 Bh Bohrium 264 | 108 Hs Hassium 265 | 109 Mt Meitnerium 266 | 110 Ds Darmstadtium 267 | 111 Rg Roentgenium 268 | 112 Cn Copernicium 269 | 113 Uut Ununtrium 270 | 114 Fl Flerovium 271 | 115 Uup Ununpentium 272 | 116 Lv Livermorium 273 | 117 Uus Ununseptium 274 | 118 Uuo Ununoctium 276 | | |
| Lanthanide Series | | 57 La Lanthanum 138.905 | 58 Ce Cerium 140.12 | 59 Pr Praseodymium 140.908 | 60 Nd Neodymium 144.24 | 61 Pm Promethium 144.913 | 62 Sm Samarium 150.36 | 63 Eu Europium 151.964 | 64 Gd Gadolinium 157.25 | 65 Tb Terbium 158.925 | 66 Dy Dysprosium 162.50 | 67 Ho Holmium 164.930 | 68 Er Erbium 167.255 | 69 Tm Thulium 168.934 | 70 Yb Ytterbium 173.054 | 71 Lu Lutetium 174.967 | | | |
| Actinide Series | | 89 Ac Actinium 227 | 90 Th Thorium 232.038 | 91 Pa Protactinium 231.036 | 92 U Uranium 238.029 | 93 Np Neptunium 237.048 | 94 Pu Plutonium 244.064 | 95 Am Americium 243.061 | 96 Cm Curium 247.070 | 97 Bk Berkelium 247.070 | 98 Cf Californium 251.080 | 99 Es Einsteinium 252.083 | 100 Fm Fermium 257.105 | 101 Md Mendelevium 258.10 | 102 No Nobelium 259.10 | 103 Lr Lawrencium 260.10 | | | |
| Alkali Metal | | Alkaline Earth | | | Transition Metal | | | Block | | Semimetal | | Nonmetal | | Halogen | | Noble Gas | | Actinide | |

Light-emitting stimuli-responsive polymer networks



Lanthanide salt

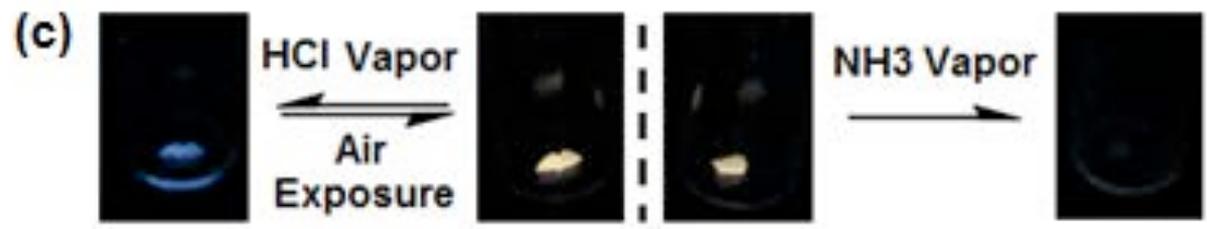
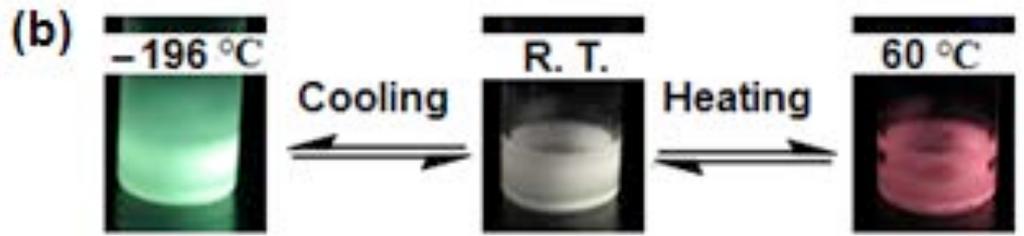
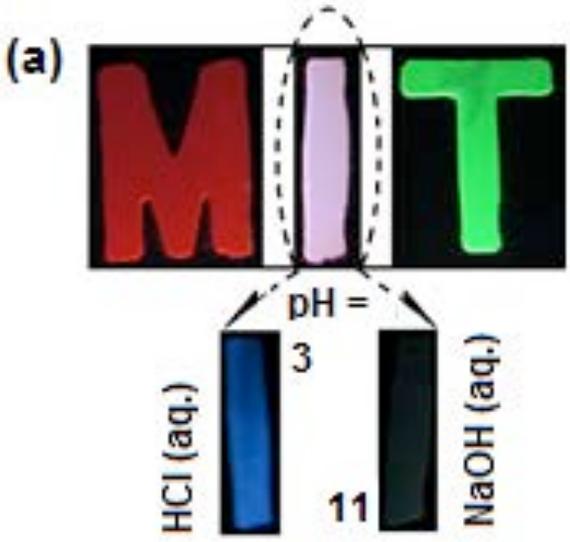
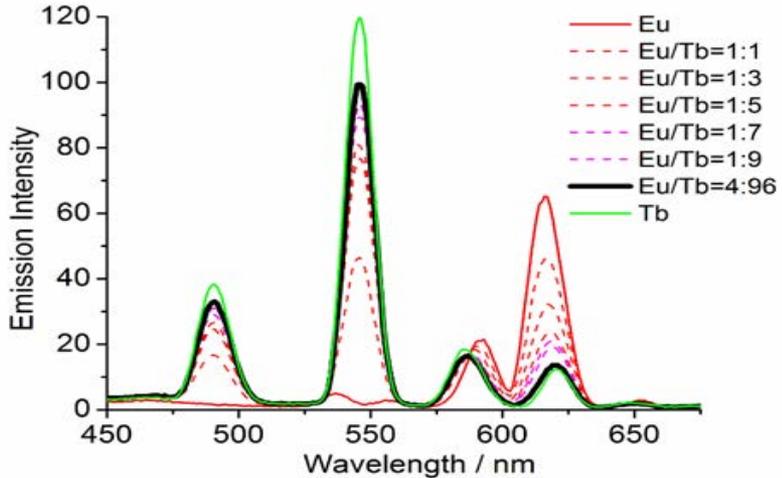
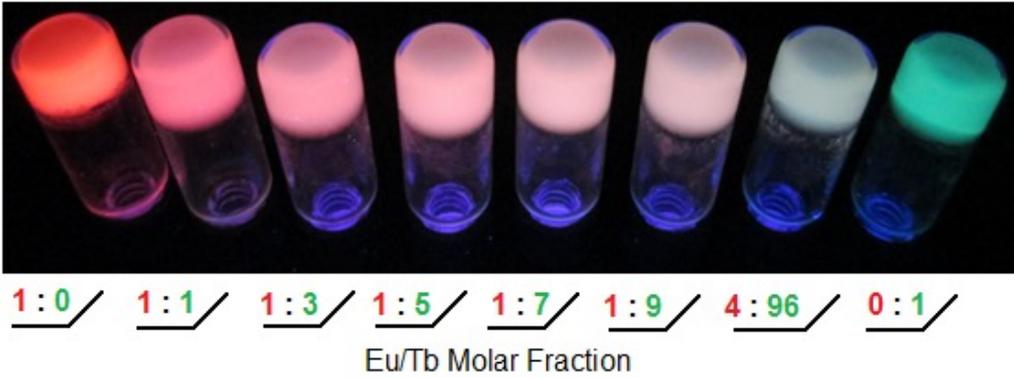


Polymer

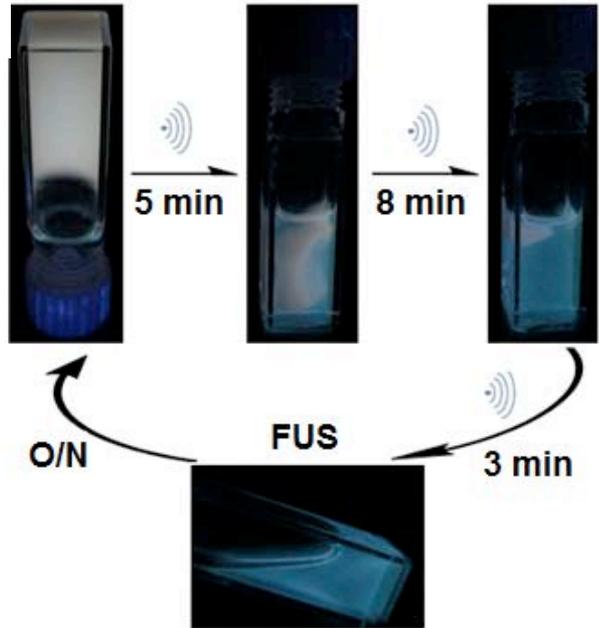
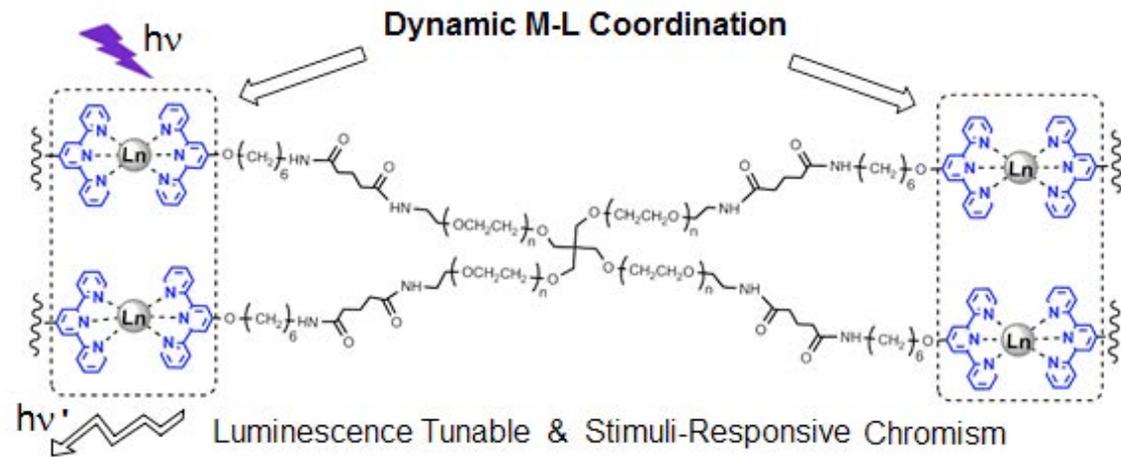


| Ln(III) = | | |
|-----------|-------|-------|
| Eu | Eu/Tb | Tb |
| | | |
| RLGel | WLGel | GLGel |

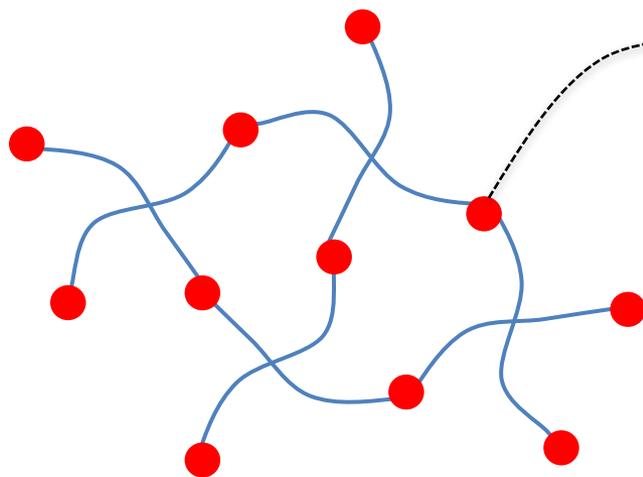
Light-emitting stimuli-responsive polymer networks



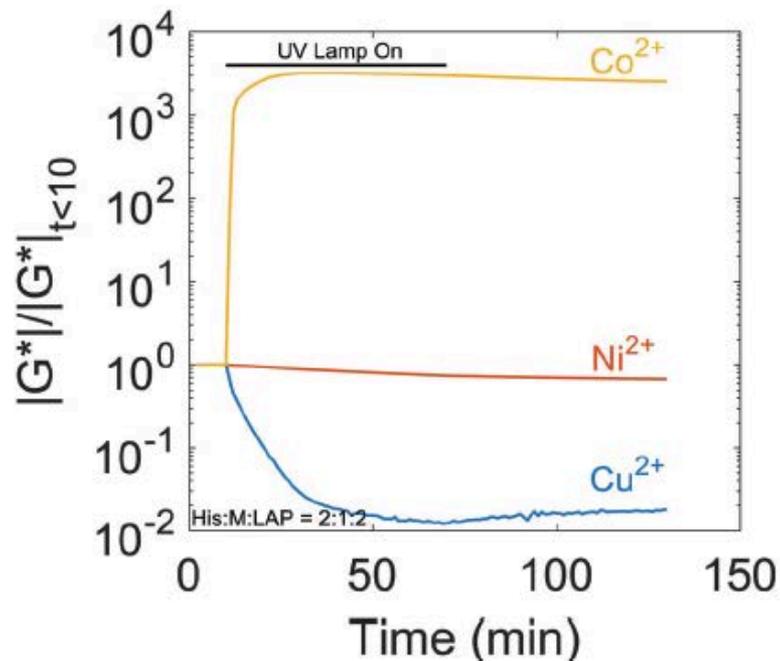
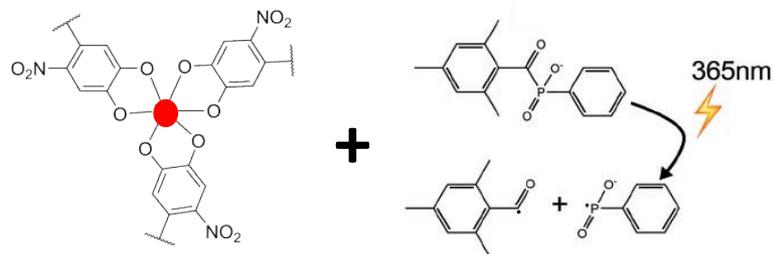
Light-emitting stimuli-responsive polymer networks



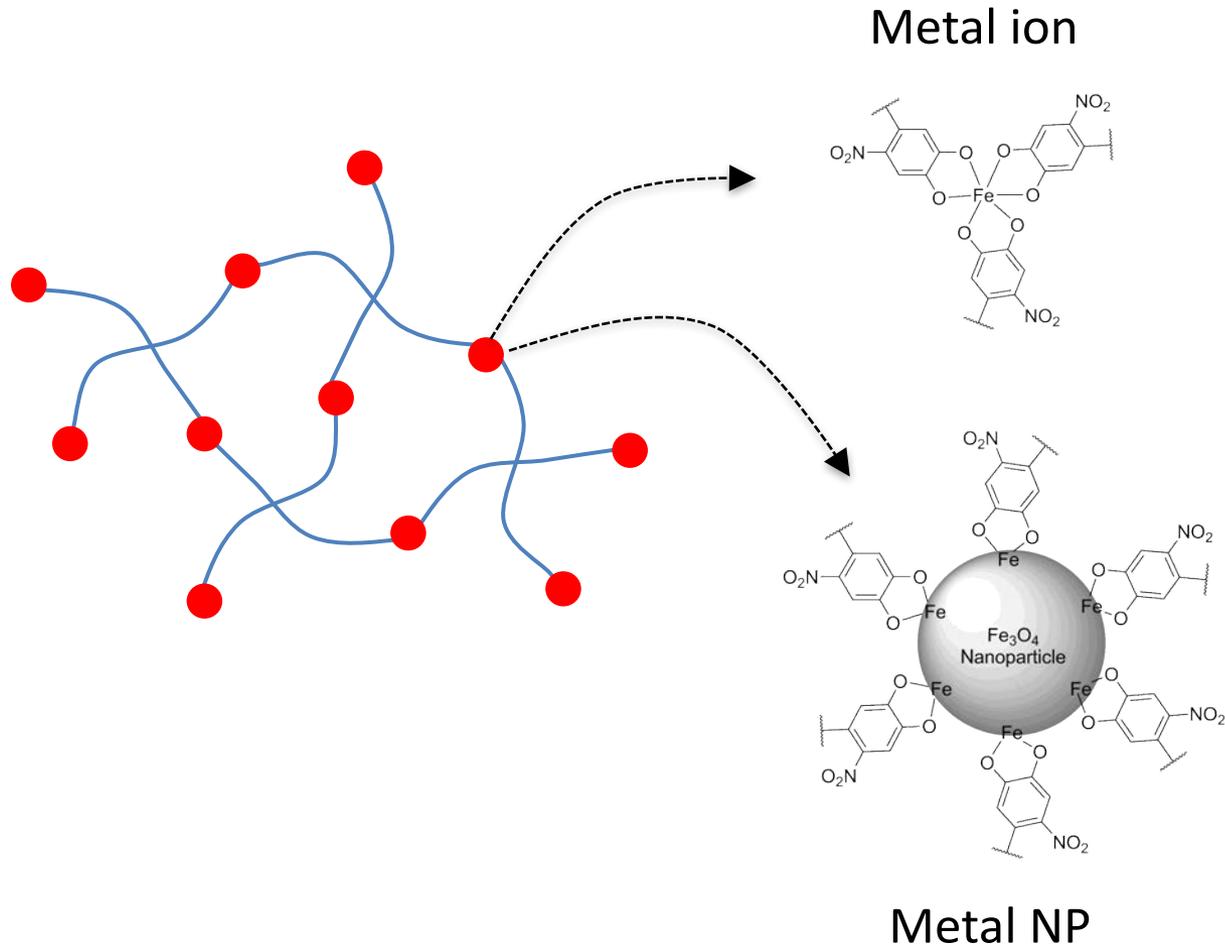
Metal ion-coordinate dynamics set gel mechanics



Metal ion

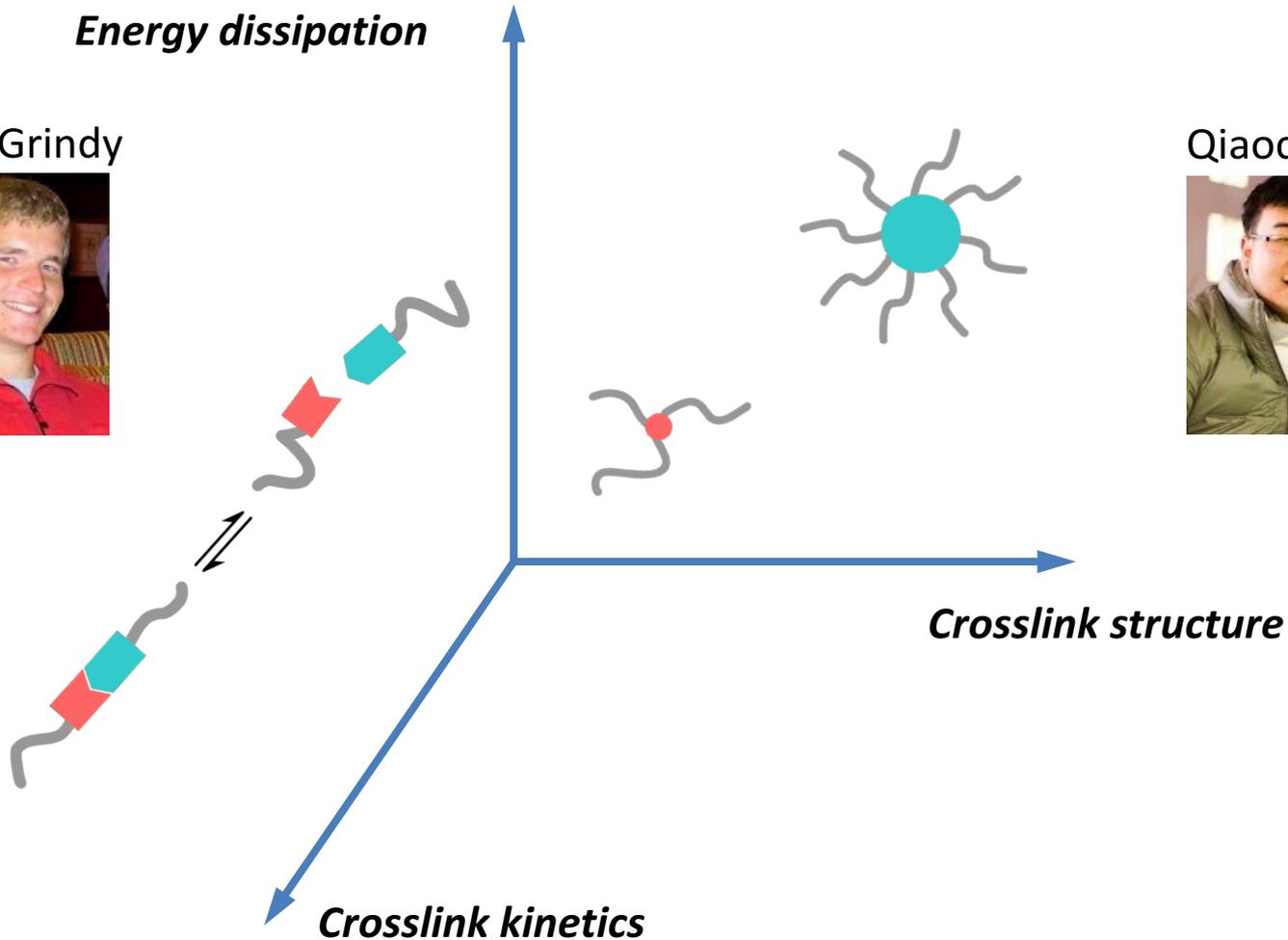


Bio-inspired metal-coordinate crosslinked networks



Energy dissipative crosslink engineering

Scott Grindy

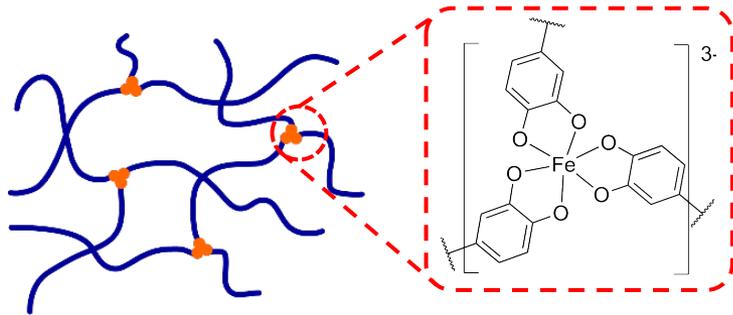


Qiaochu Li

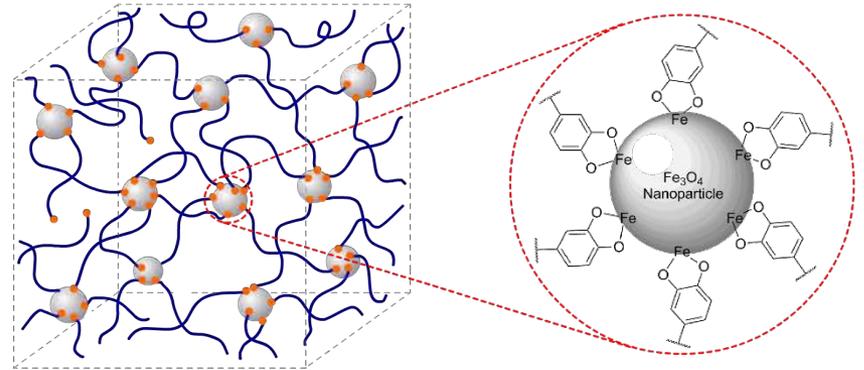


Energy dissipative crosslinks: Ions vs Nano-particles

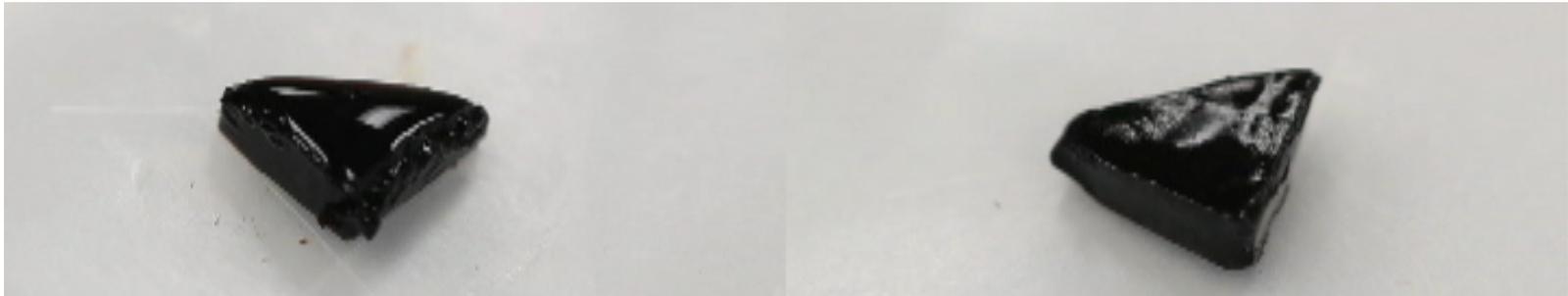
f : number of ligands at each crosslink



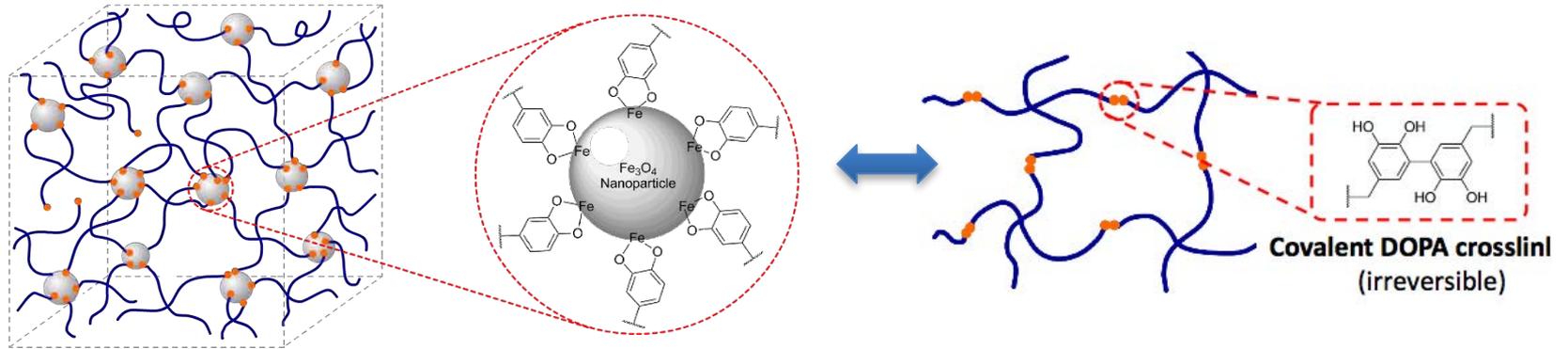
$$f = 3$$



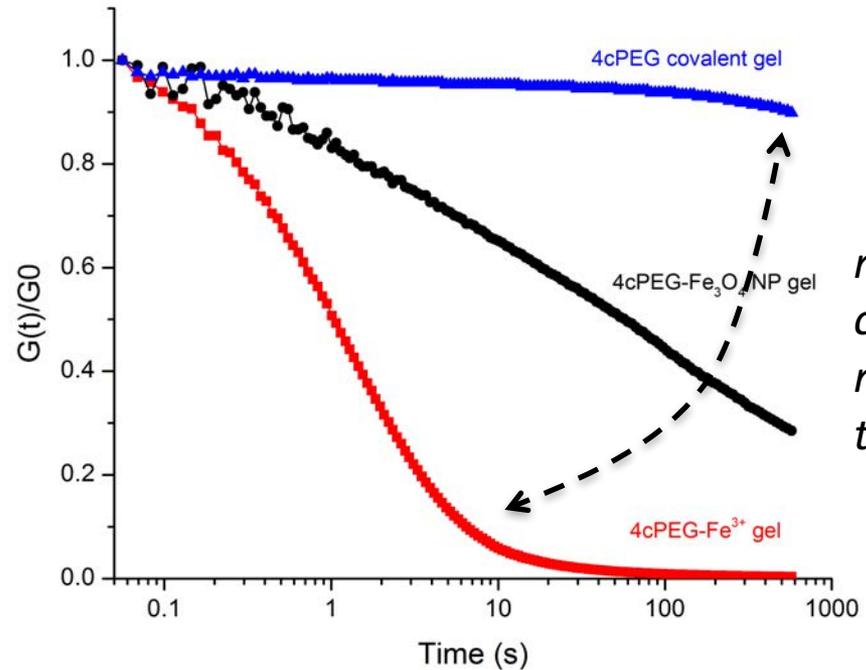
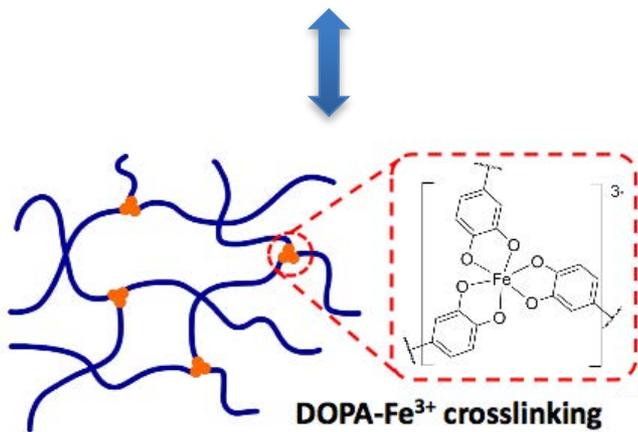
$$f \sim 10^2$$



Energy dissipative crosslinks: Ions vs Nano-particles



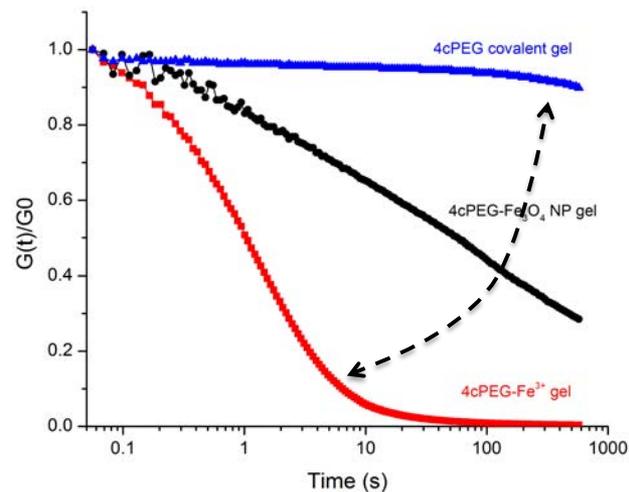
Self-assembled Hydrogel Network



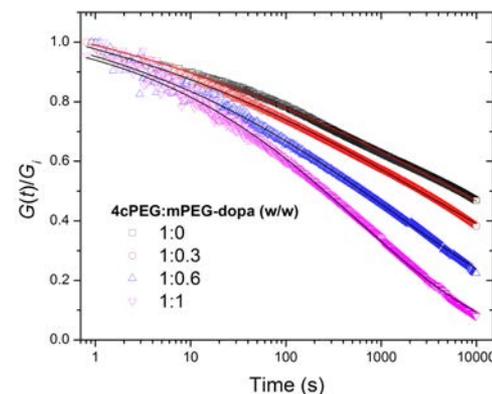
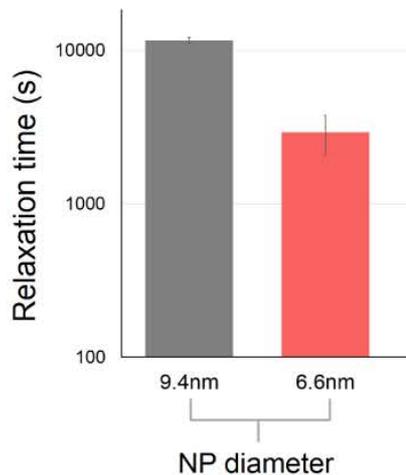
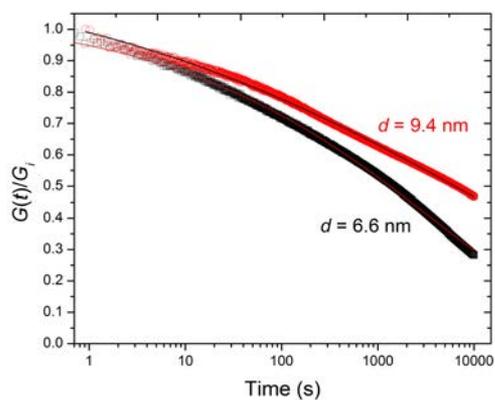
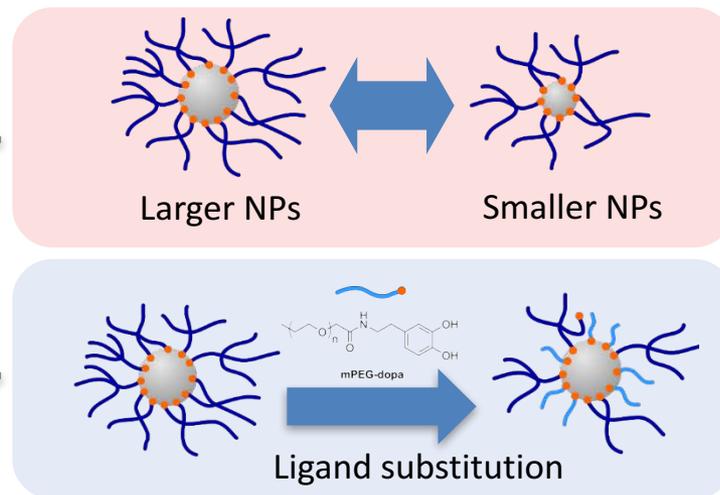
role of f in controlling relaxation time τ ?

Energy dissipative crosslinks:

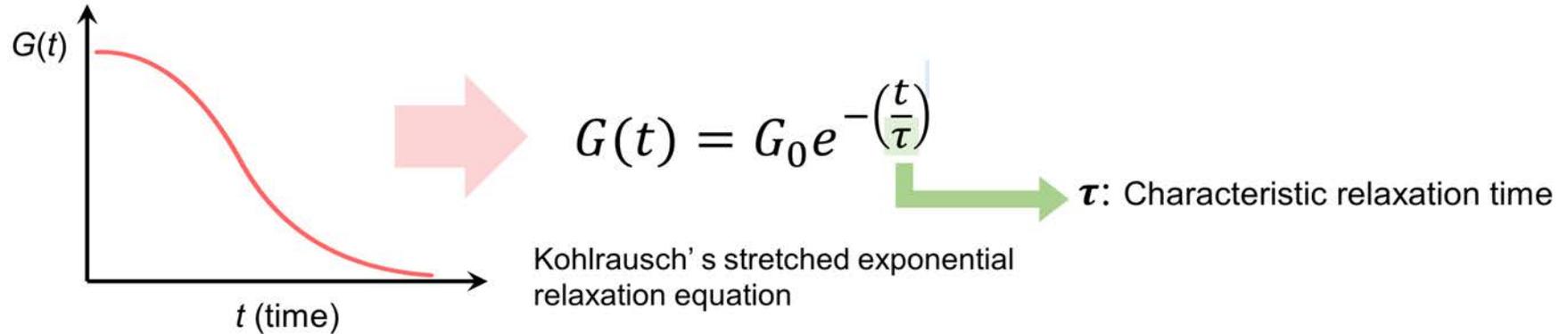
$$\tau \propto f$$



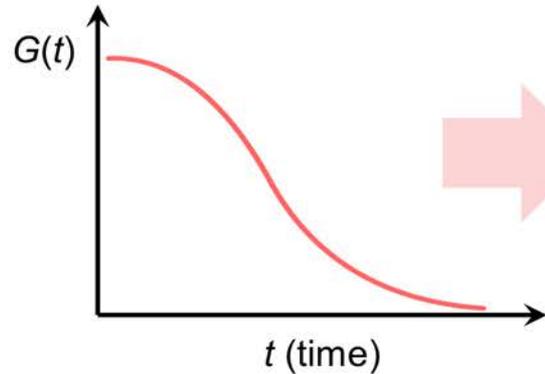
$$\tau \propto f$$



Energy dissipative crosslinks: $\tau \propto f$



Energy dissipative crosslinks: α

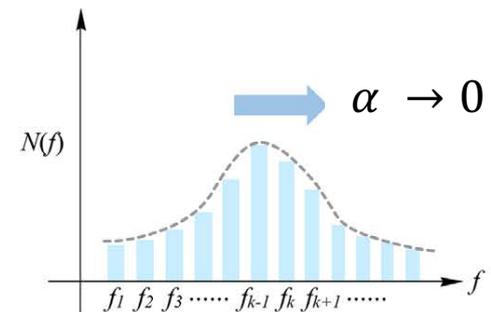
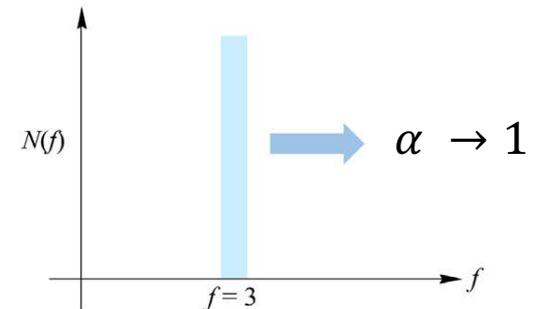


$$G(t) = G_0 e^{-\left(\frac{t}{\tau}\right)^\alpha}$$

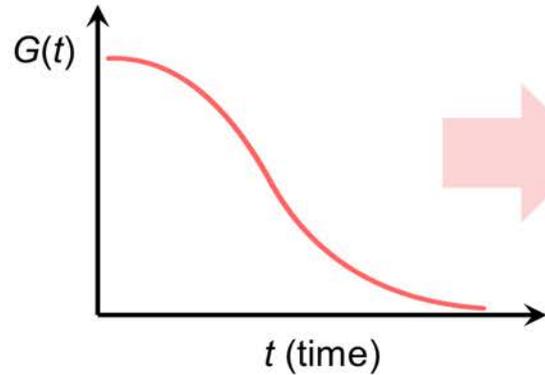
α : Kohlrausch's exponent

τ : Characteristic relaxation time

Kohlrausch's stretched exponential relaxation equation



Energy dissipative crosslinks: α

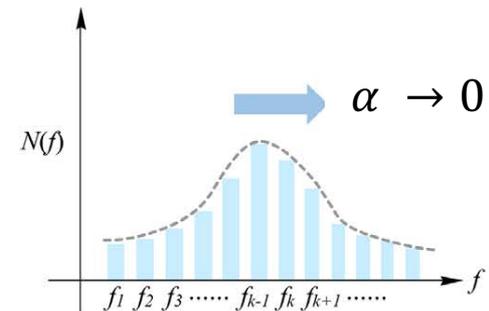
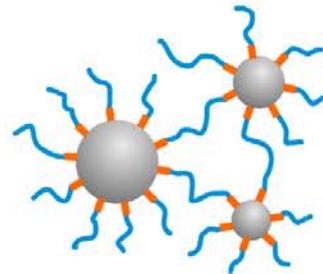
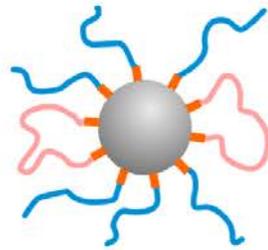
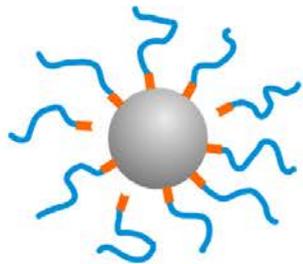
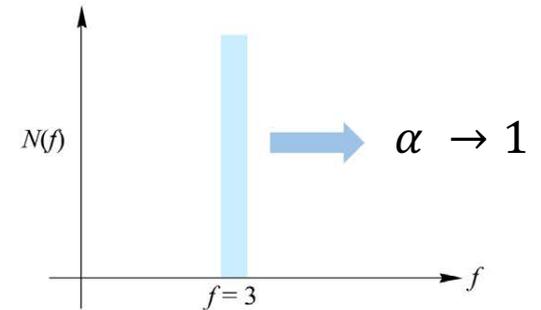
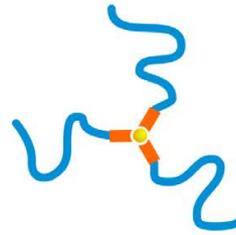


$$G(t) = G_0 e^{-\left(\frac{t}{\tau}\right)^\alpha}$$

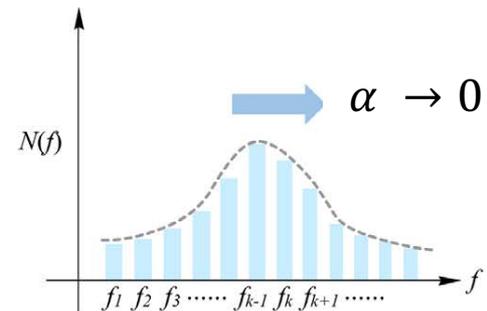
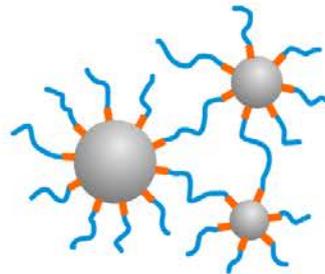
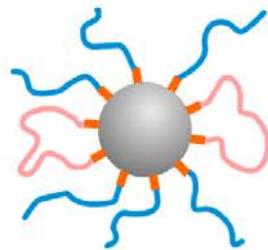
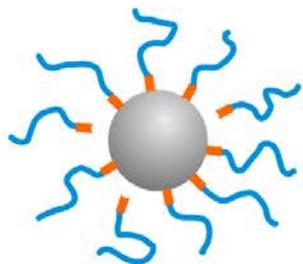
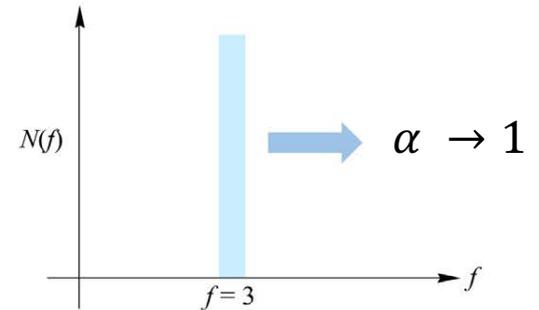
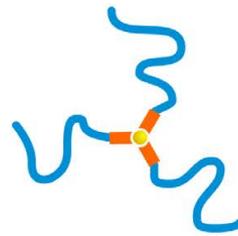
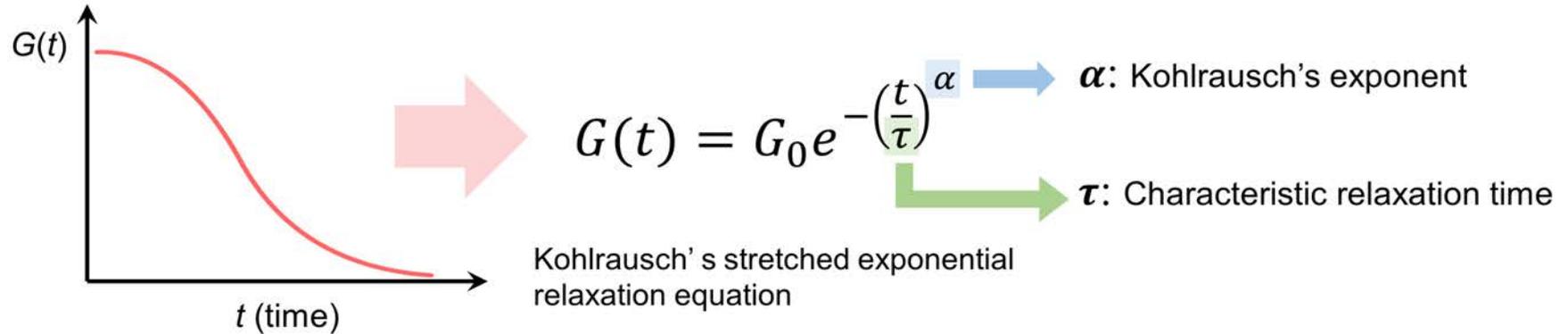
α : Kohlrausch's exponent

τ : Characteristic relaxation time

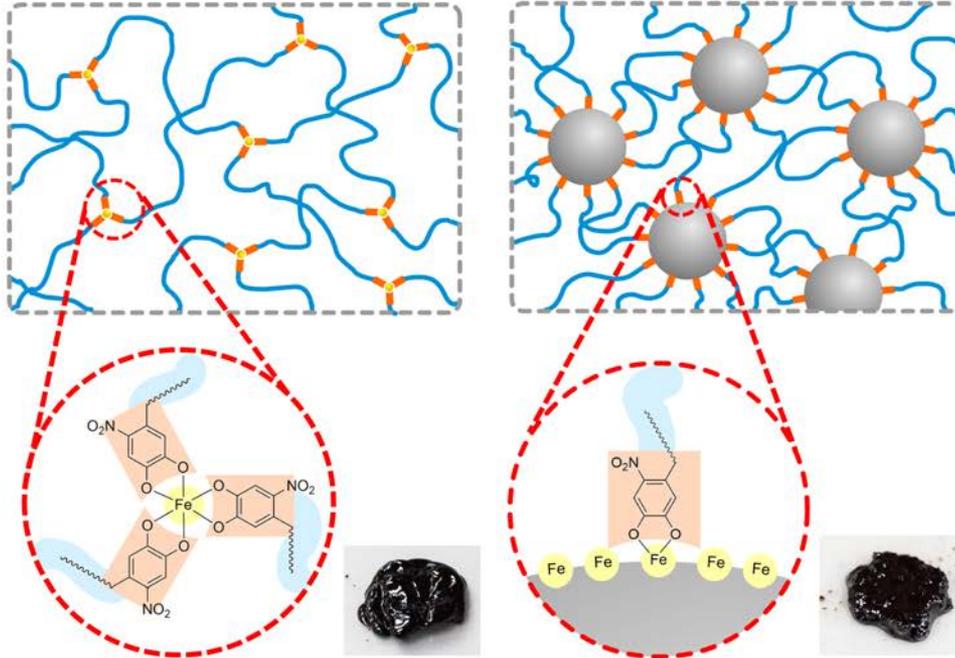
Kohlrausch's stretched exponential relaxation equation



Energy dissipative crosslinks: $\alpha \propto 1/(\Delta f)$

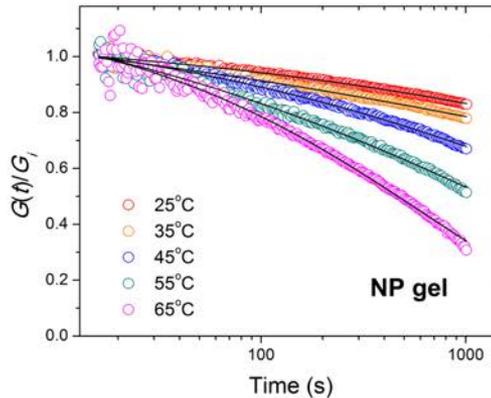
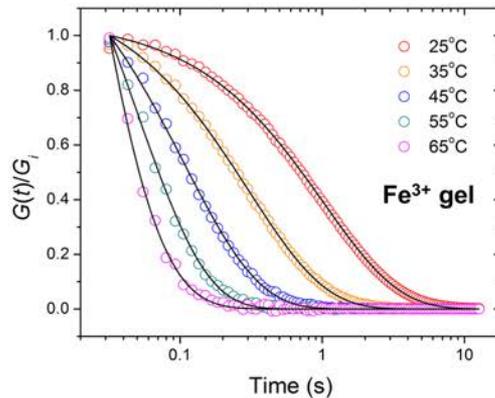


Energy dissipative crosslinks : tau vs alpha

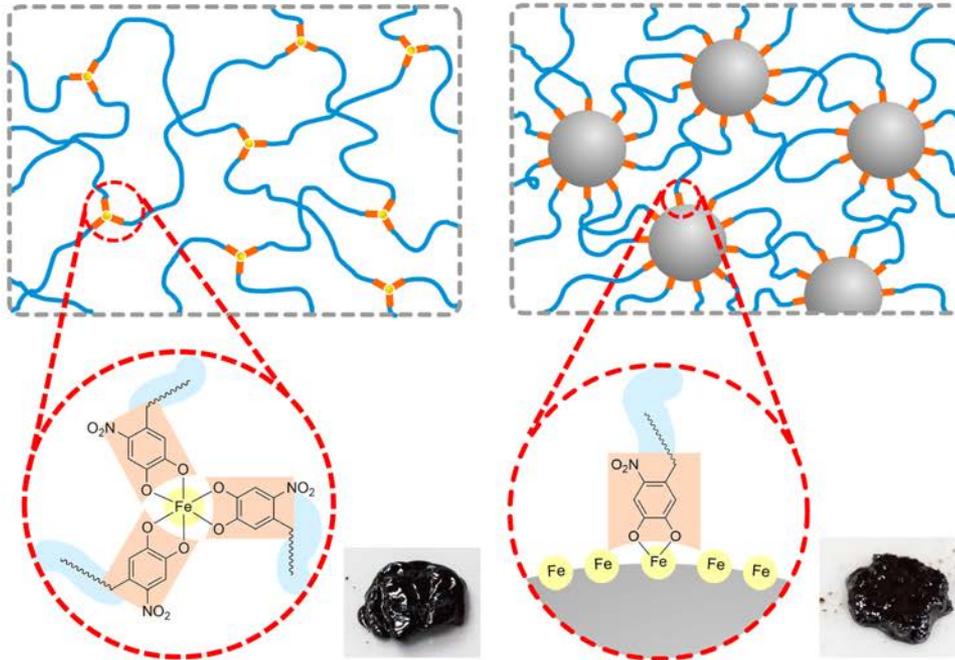


Fe³⁺ gel: ML₃ crosslink

NP gel: NP-L_n crosslink

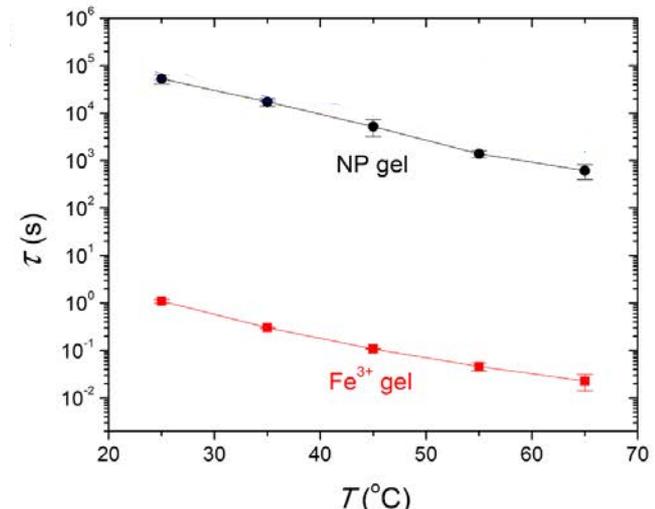
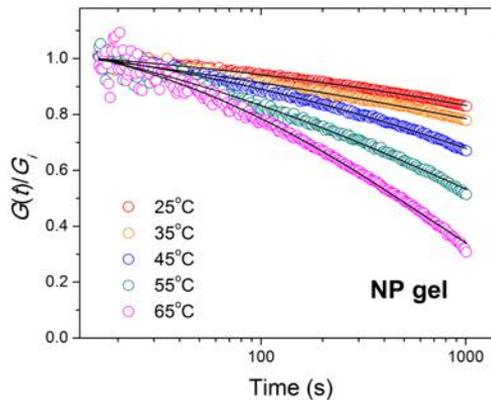
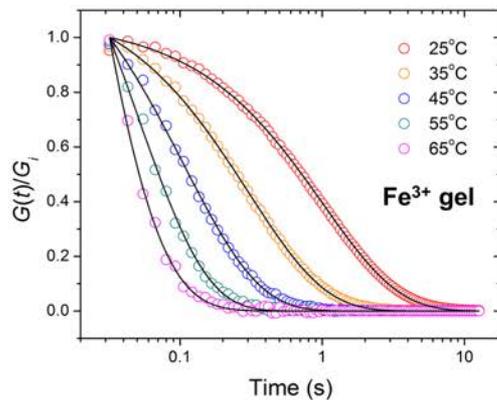


Energy dissipative crosslinks : tau vs alpha

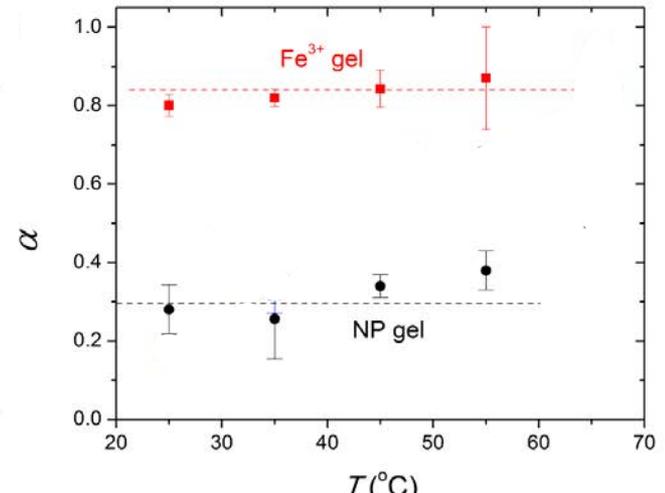
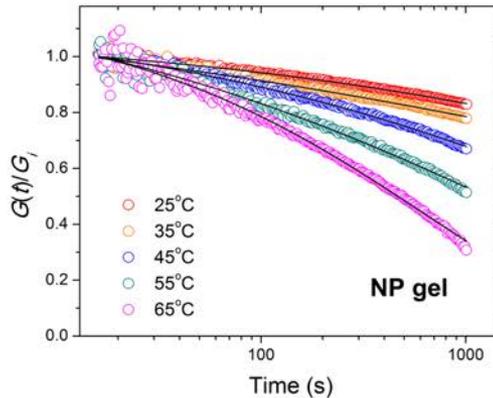
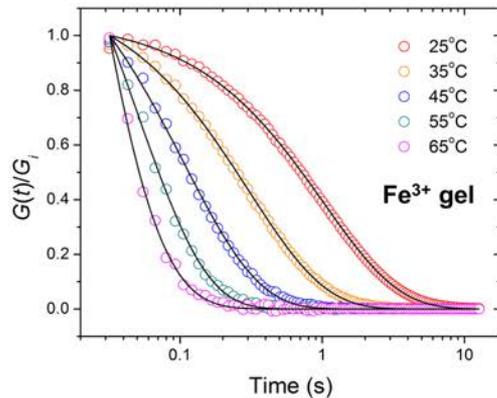
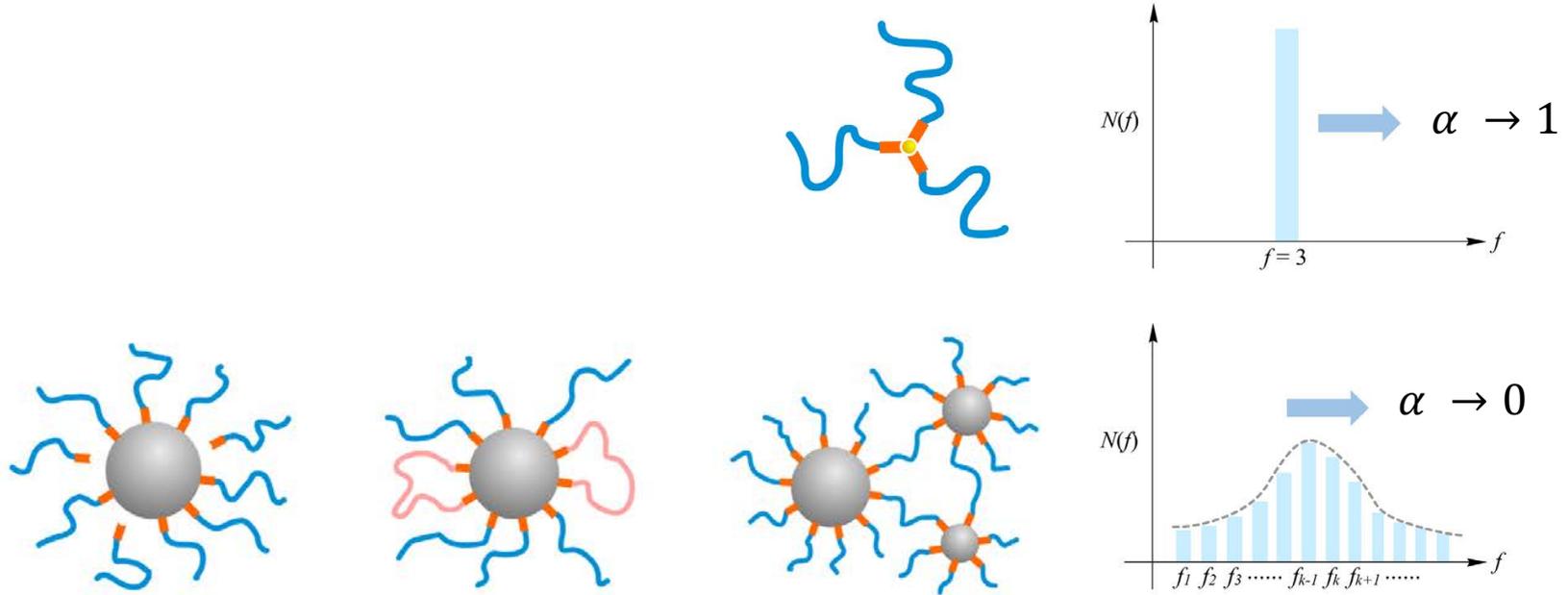


Fe³⁺ gel: ML₃ crosslink

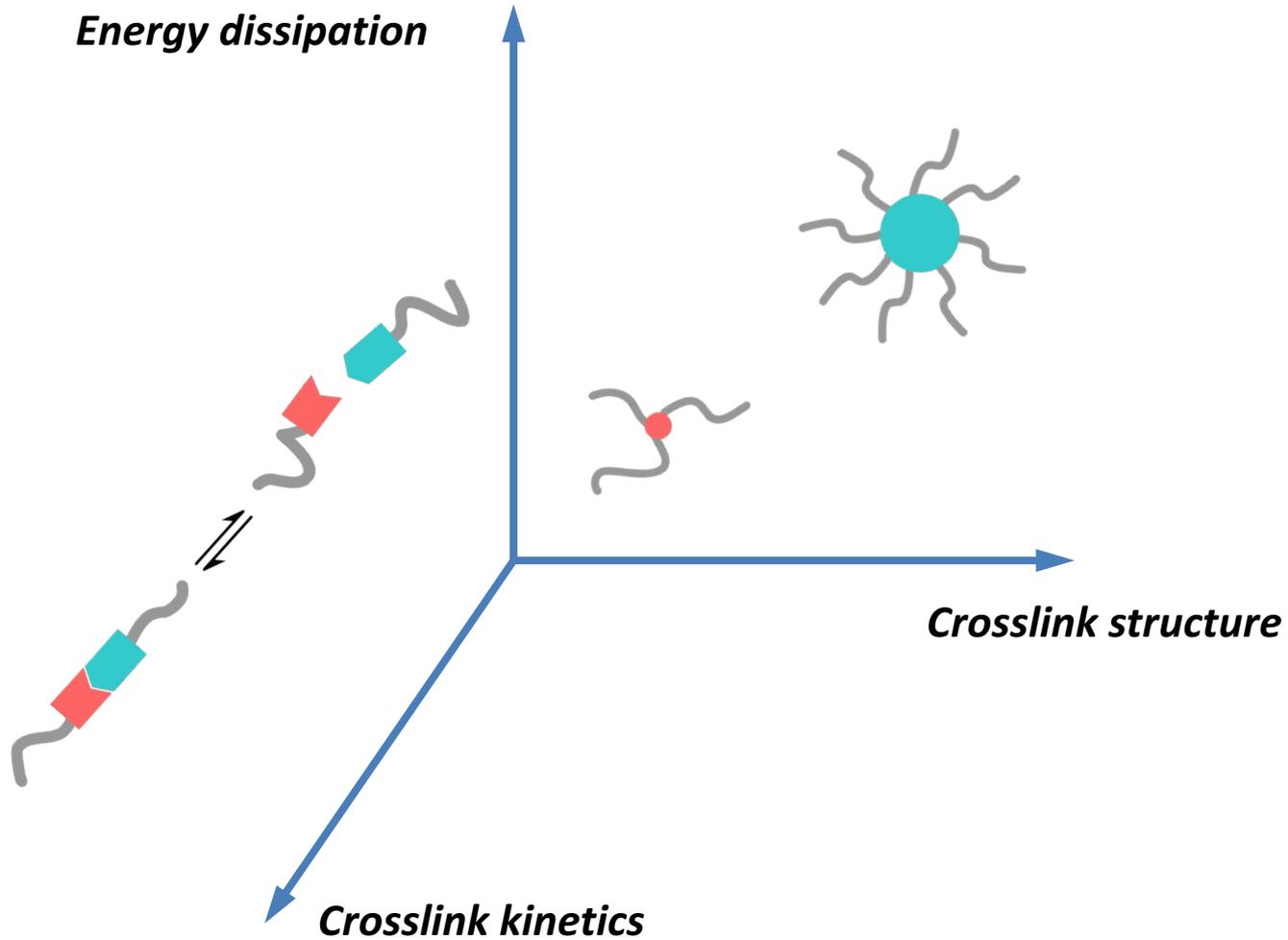
NP gel: NP-L_n crosslink



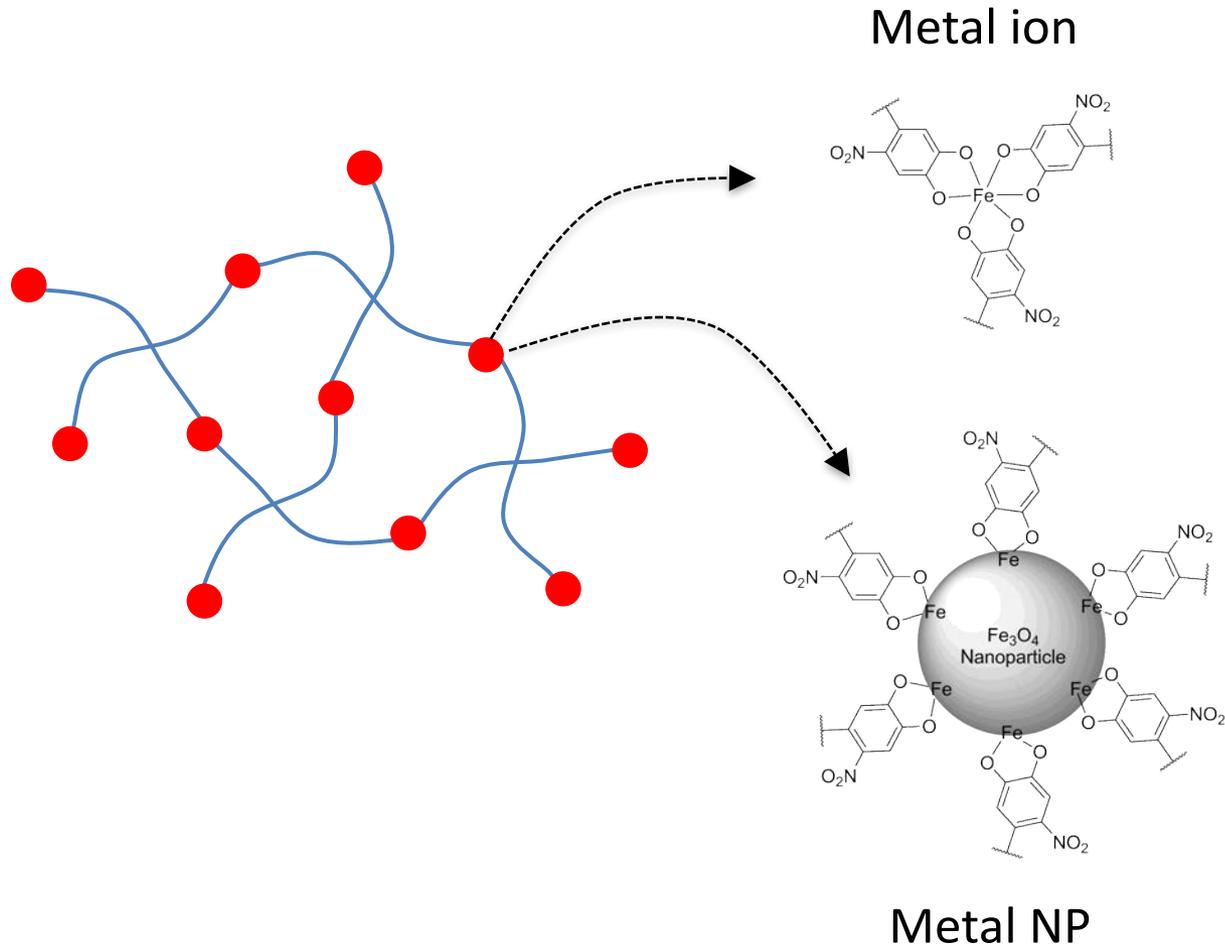
Energy dissipative crosslinks : tau vs alpha



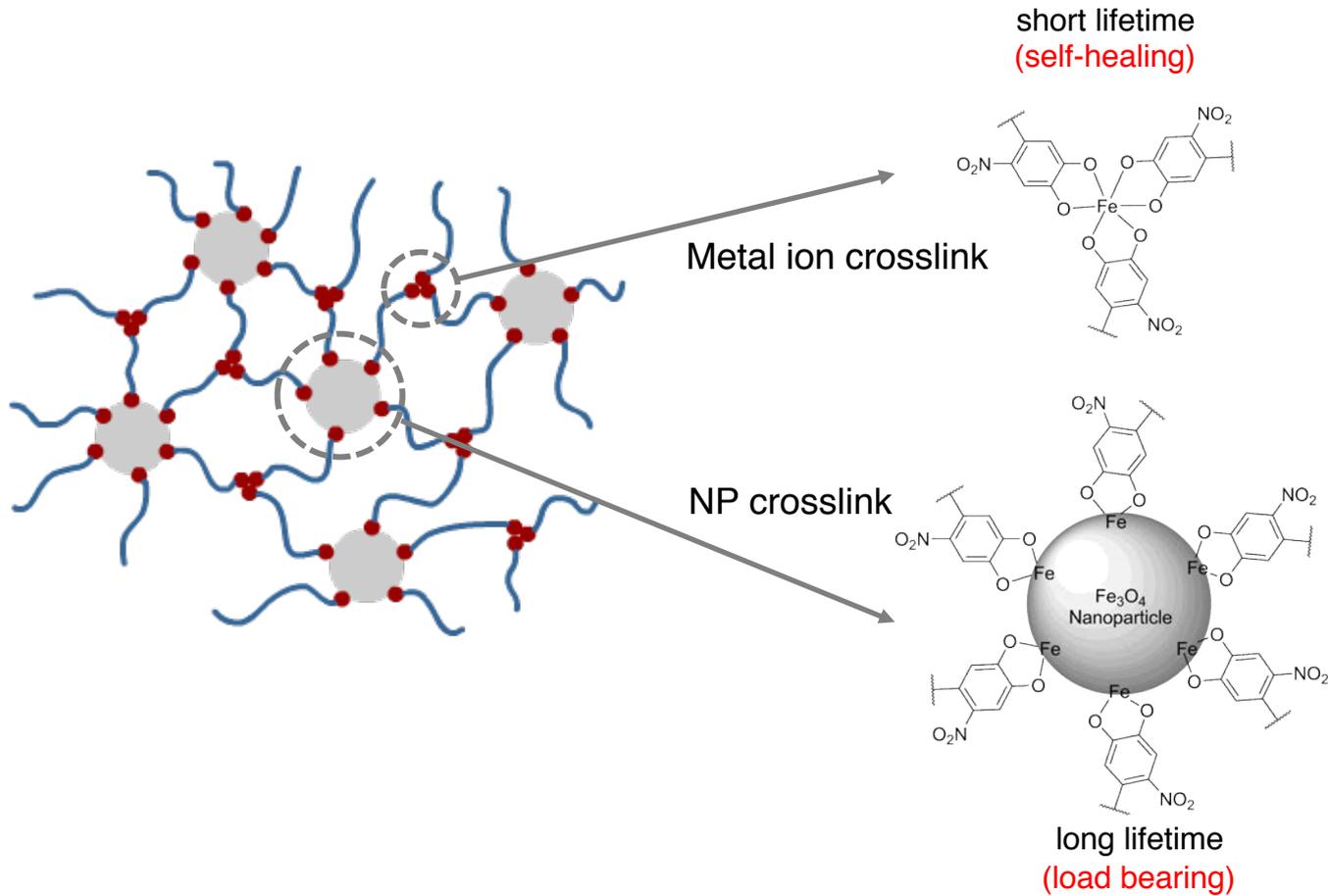
Energy dissipative crosslink engineering



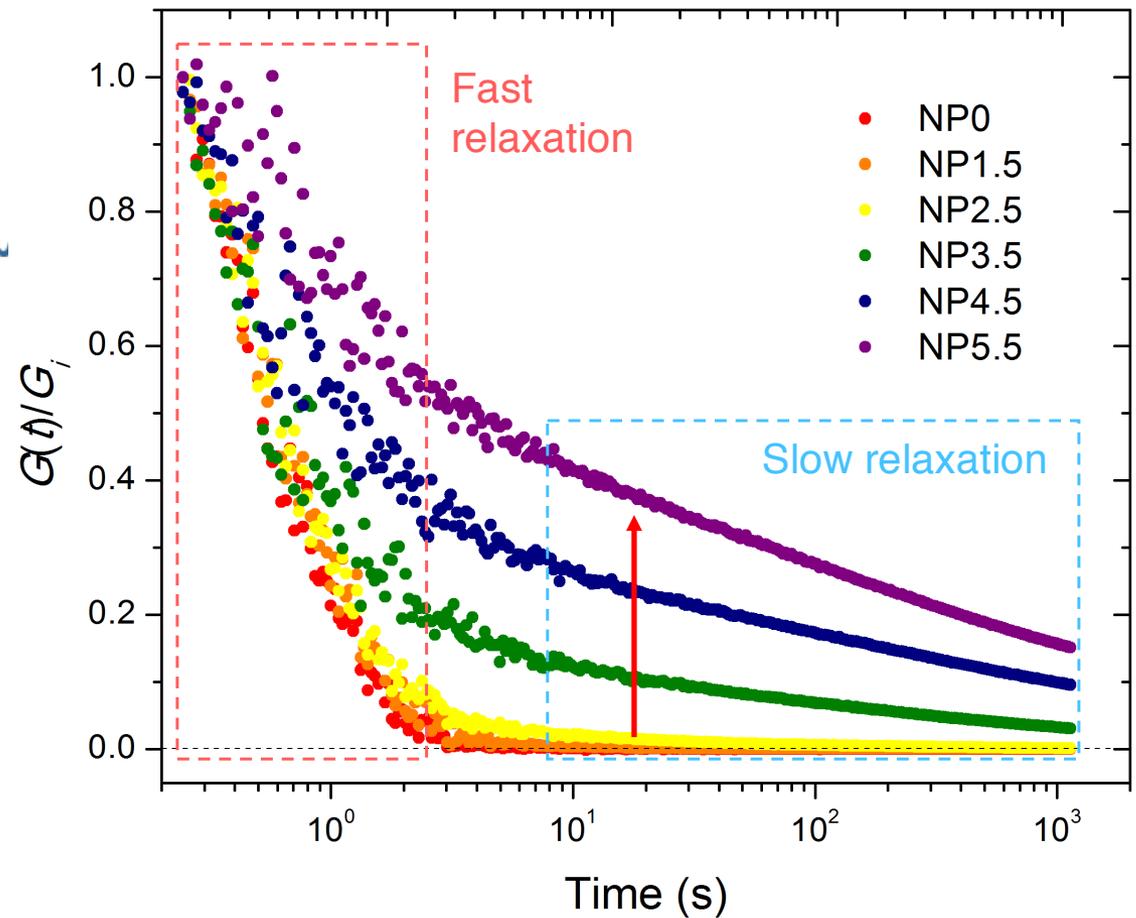
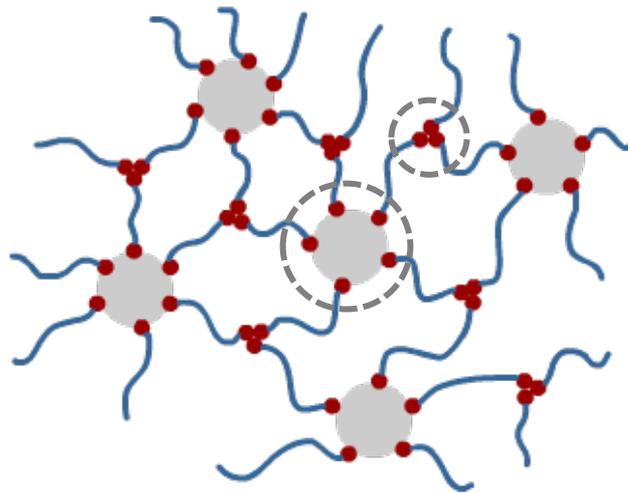
Energy dissipative crosslink engineering



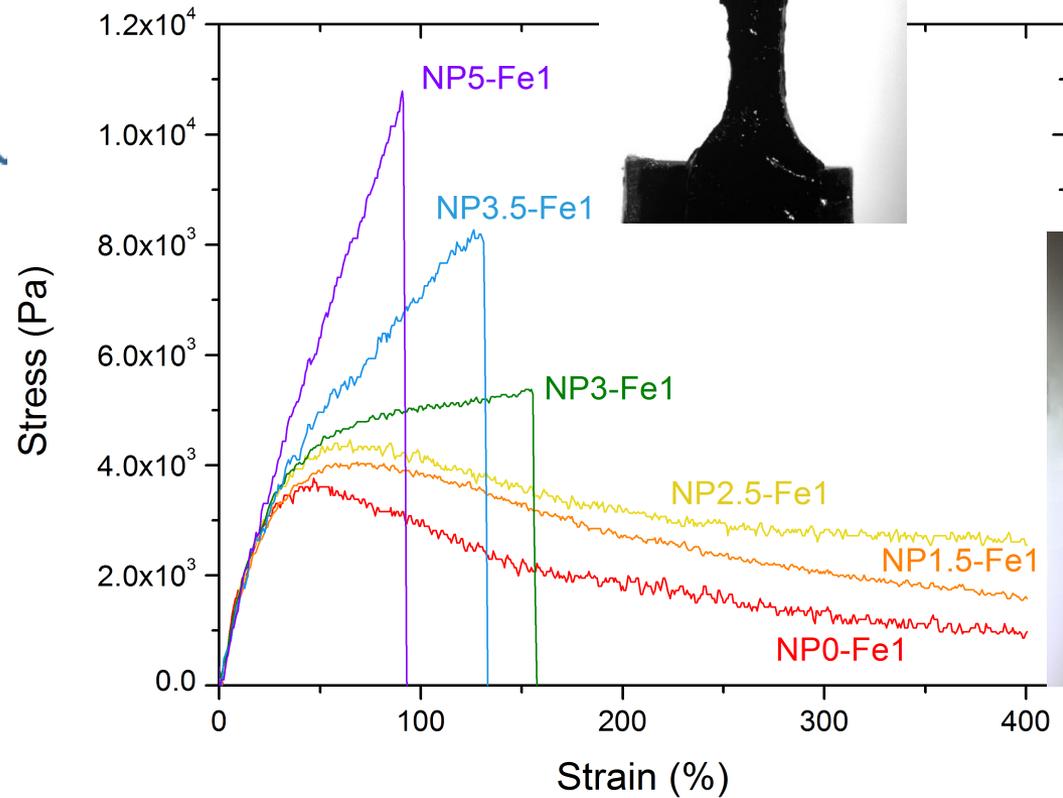
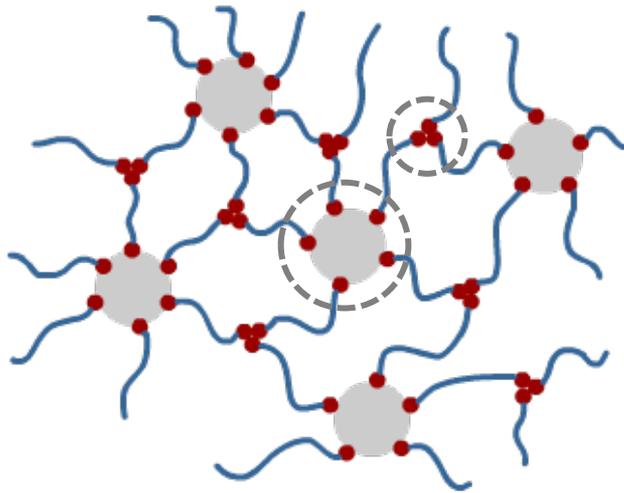
Energy dissipative crosslink engineering

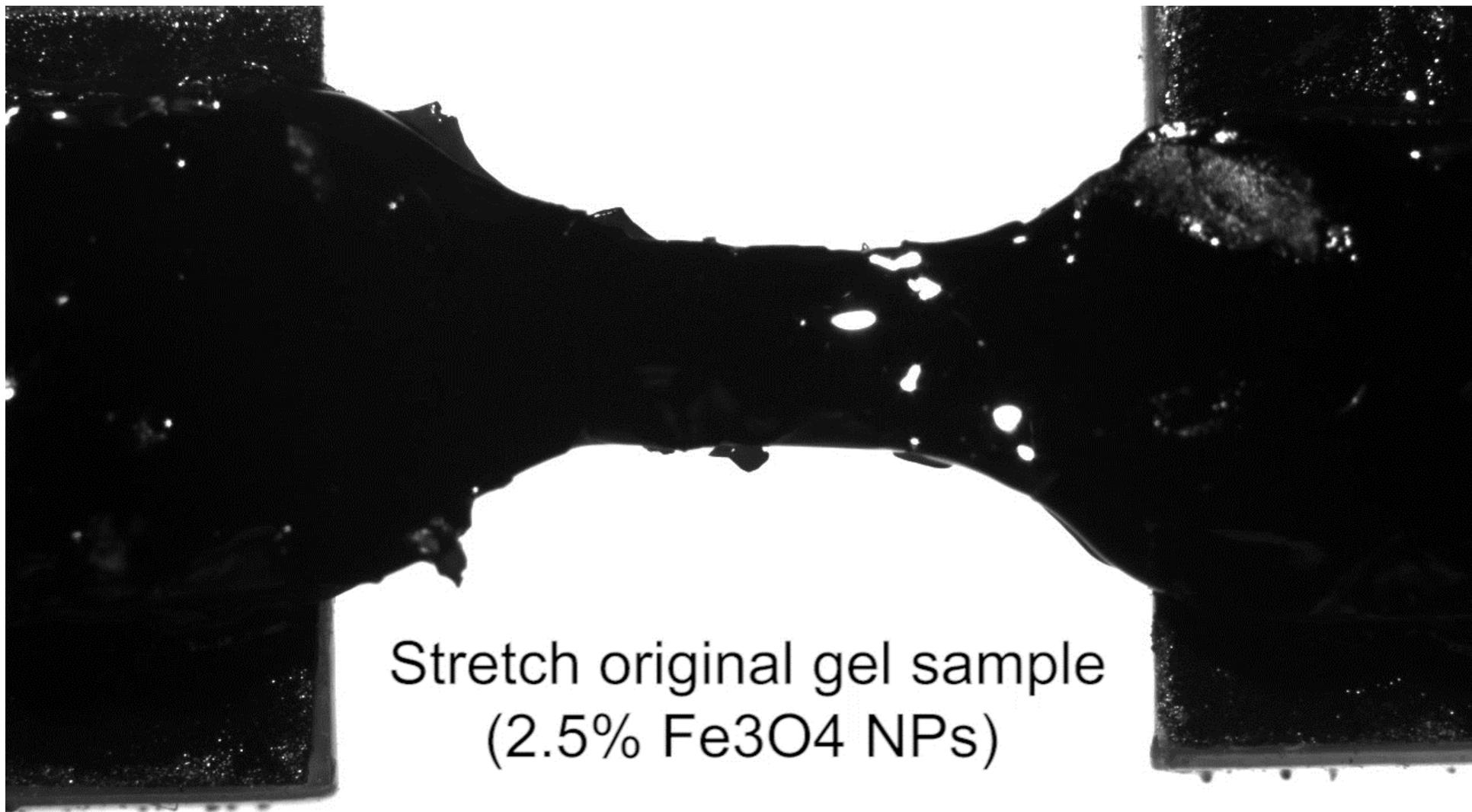


Energy dissipative crosslink engineering



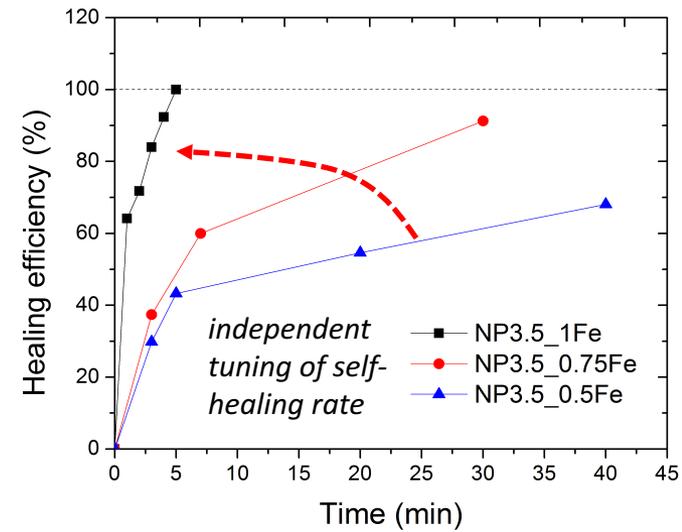
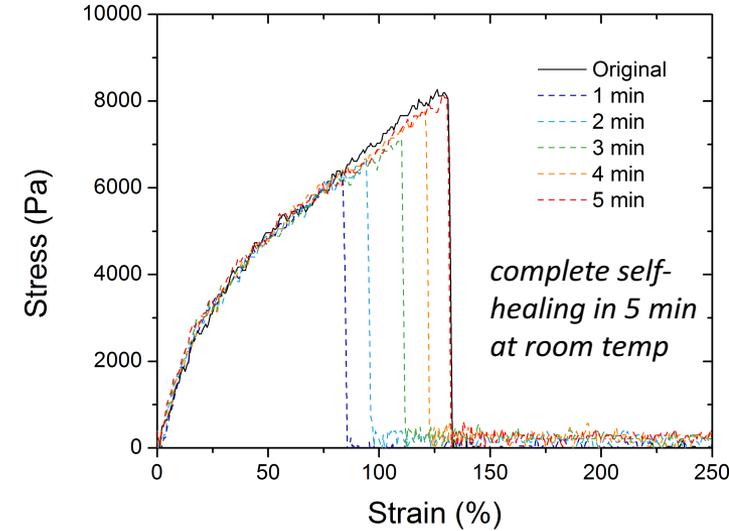
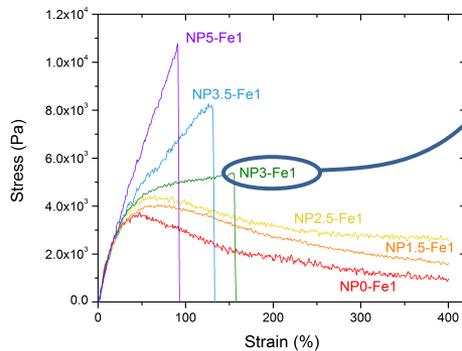
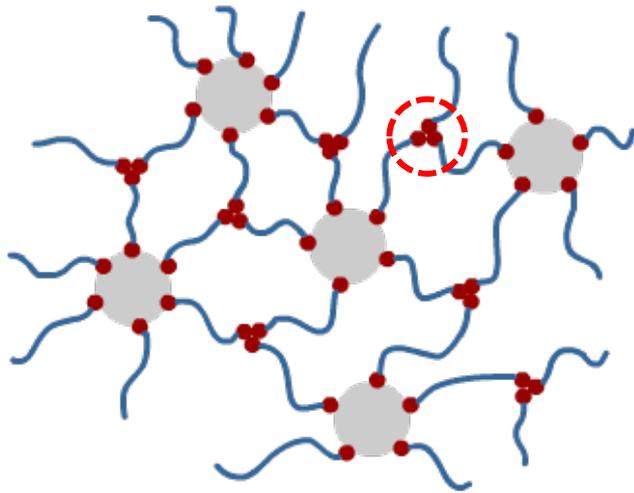
Energy dissipative crosslink engineering



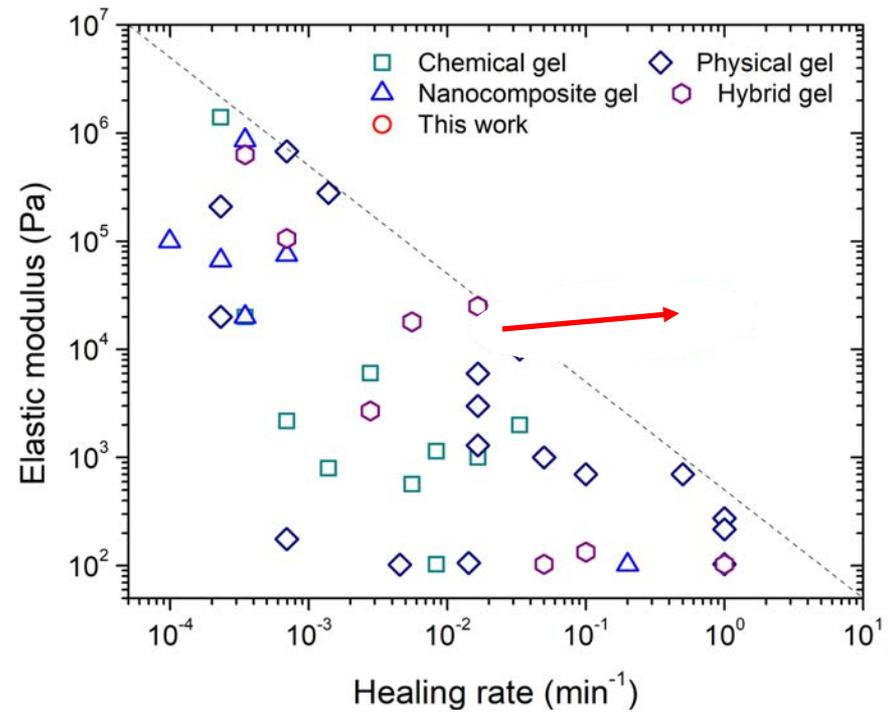
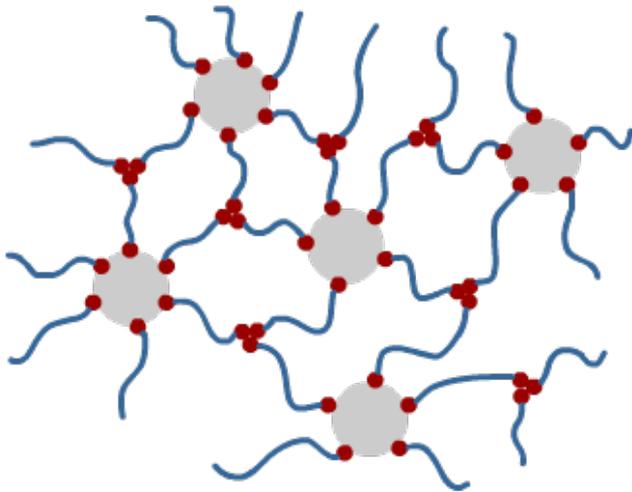


Stretch original gel sample
(2.5% Fe₃O₄ NPs)

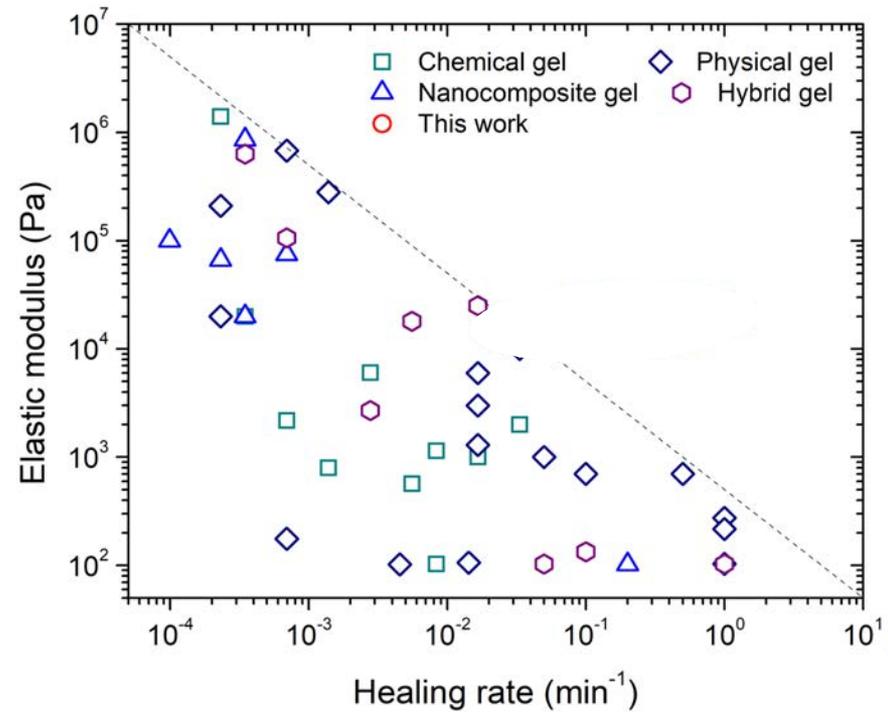
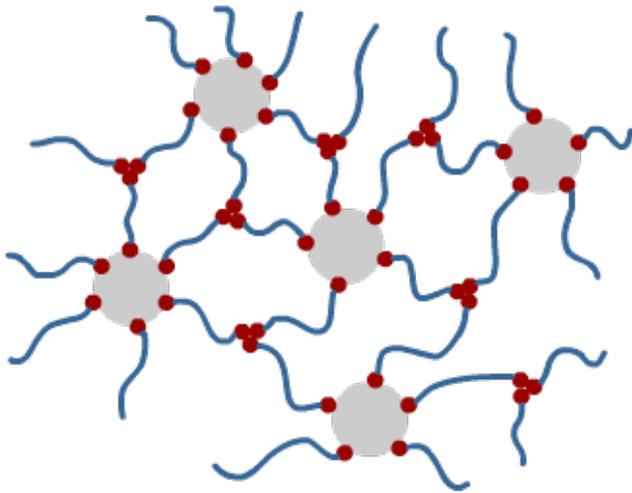
Energy dissipative crosslink engineering



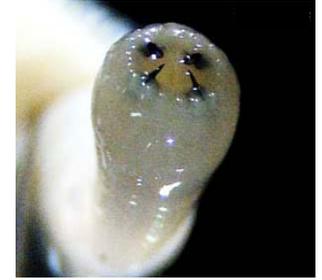
Energy dissipative crosslink engineering



Energy dissipative crosslink engineering

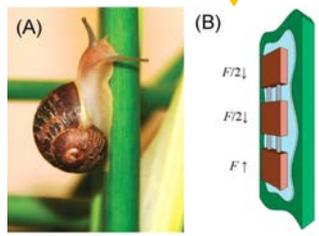


More inspiration.....

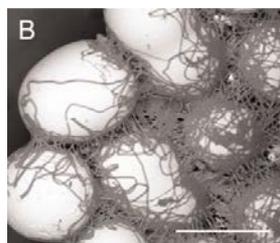
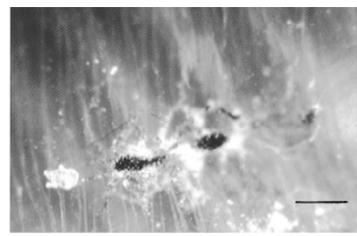


underwater sealants

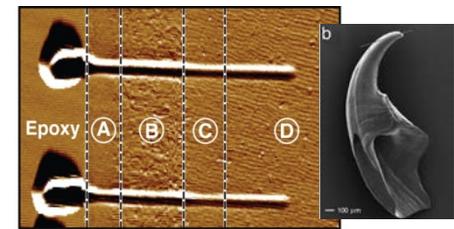
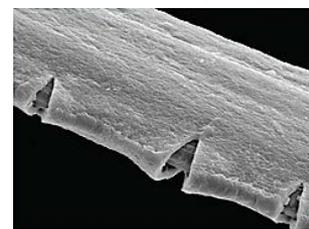
self-healing fibers



reversible adhesives



underwater adhesives



light-weight high hardness materials

More inspiration.....



Visco-elastic fluids

Hard solids

More inspiration.....



REPORTS

MATERIALS SCIENCE

Toughening elastomers using mussel-inspired iron-catechol complexes

Emmanouela Filippidi,^{1,2*} Thomas R. Cristiani,^{1,3*} Claus D. Eisenbach,^{1,4} J. Herbert Waite,^{1,5} Jacob N. Israelachvili,^{1,3,6} B. Kollbe Ahn,⁷ Megan T. Valentine^{1,2†}

Materials often exhibit a trade-off between stiffness and extensibility; for example, strengthening elastomers by increasing their cross-link density leads to embrittlement and decreased toughness. Inspired by cuticles of marine mussel byssi, we circumvent this inherent trade-off by incorporating sacrificial, reversible iron-catechol cross-links into a dry, loosely cross-linked epoxy network. The iron-containing network exhibits two to three orders of magnitude increases in stiffness, tensile strength, and tensile toughness compared to its iron-free precursor while gaining recoverable hysteretic energy dissipation and maintaining its original extensibility. Compared to previous realizations of this chemistry in hydrogels, the dry nature of the network enables larger property enhancement owing to the cooperative effects of both the increased cross-link density given by the reversible iron-catecholate complexes and the chain-restricting ionic nanodomains that they form.

Visco-elastic fluids

Hard solids



THANK YOU

Collaborators:

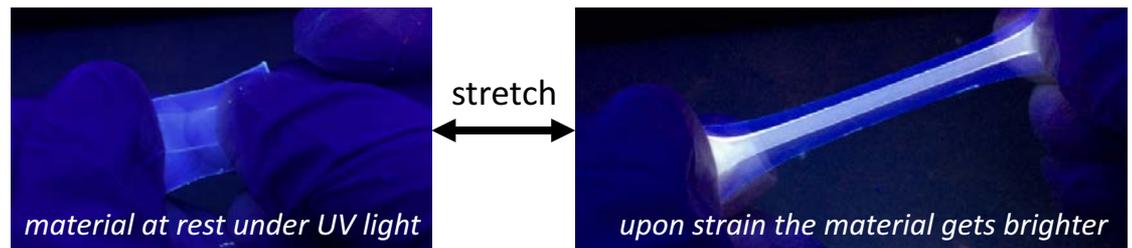
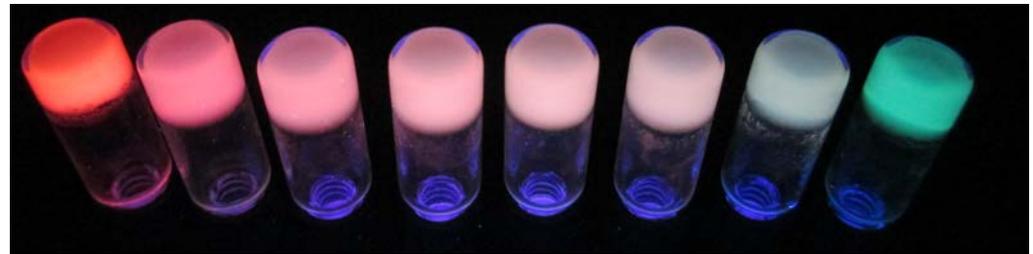
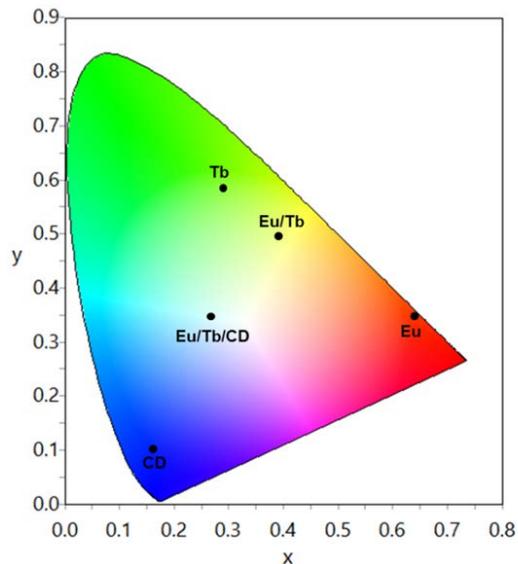
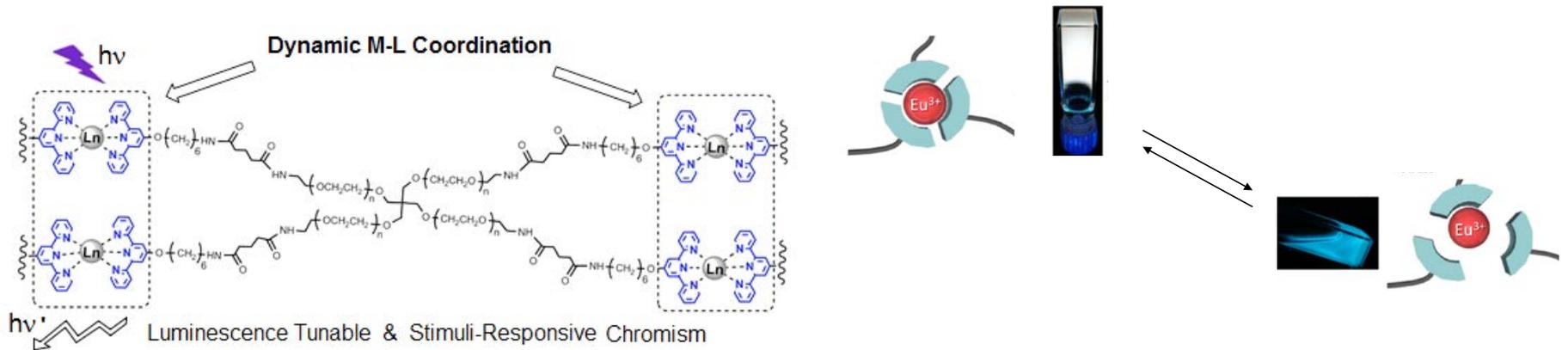
Gareth McKinley
(MIT)

Joe Tracy
(NCSU)

Grindy *et al*, Nature Materials (2015)
Chen *et al*, Adv. Opt. Mat. (2015)
Chen *et al*, JACS (2015)
Li *et al*, ACS Nano (2016)
Grindy *et al*, Macromolecules (2016)
Grindy *et al*, Soft Matter (2017)

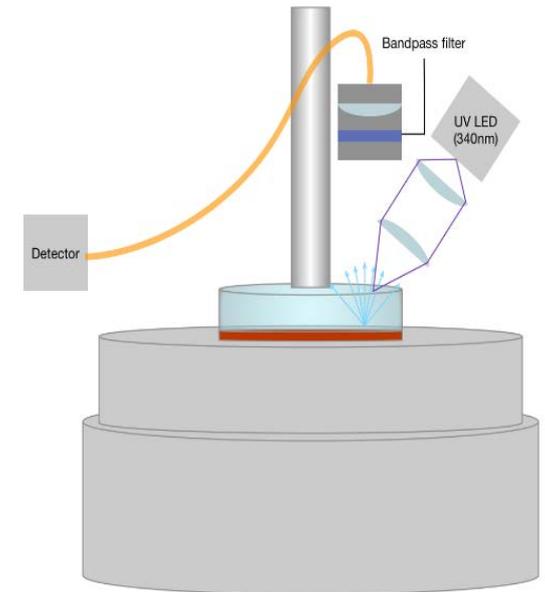
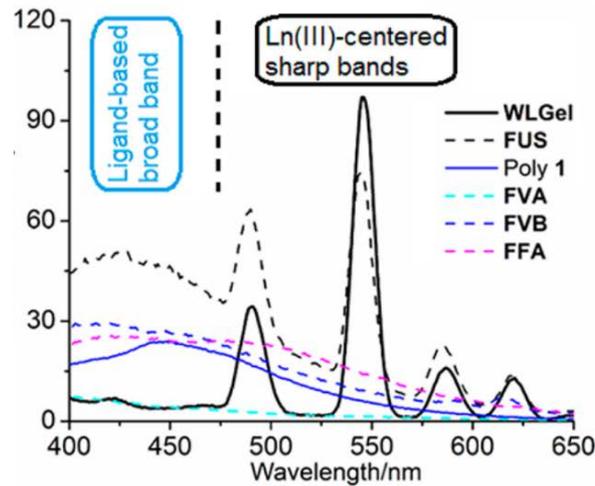
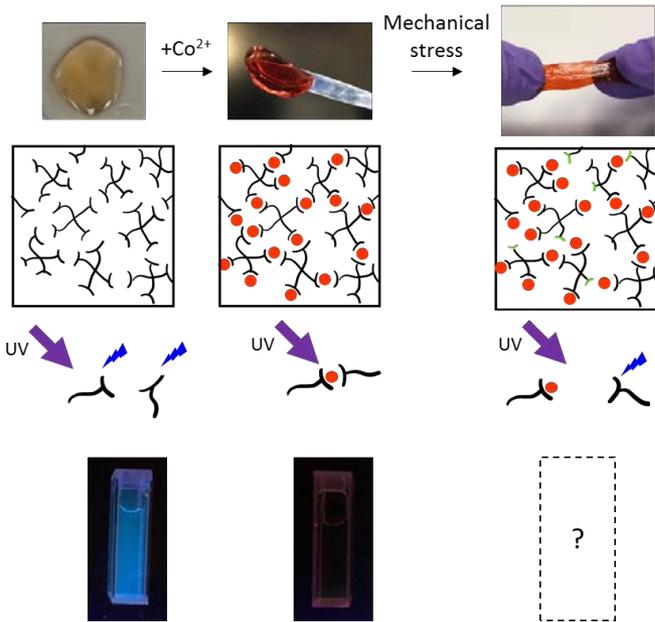
Energy dissipative crosslink engineering

To engineer self-reporting polymer materials with energy-dissipation and material failure sensing capacity: **on a macroscopic scale**



Energy dissipative crosslink engineering

To engineer self-reporting polymer materials with energy-dissipation and material failure sensing capacity: *on a microscopic scale*



Metal-coordinate networks: Ions vs particles

$$G(t) = G_0 \exp\left[-(t/\tau)^a\right], 0 < a < 1$$

$$\ln[\tau(T)] = \ln \tau_0 + \frac{E_a}{kT}$$

(pH=4)

NP gel:

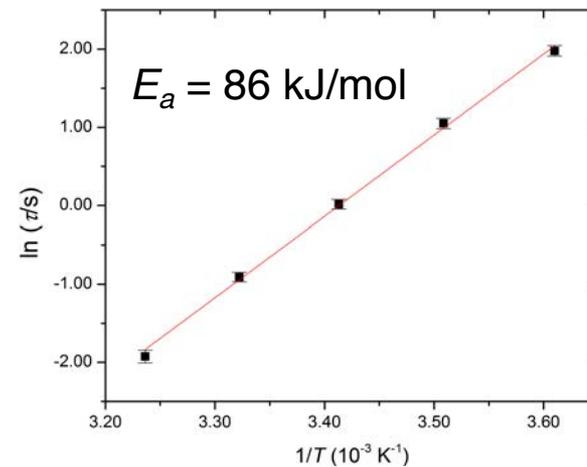
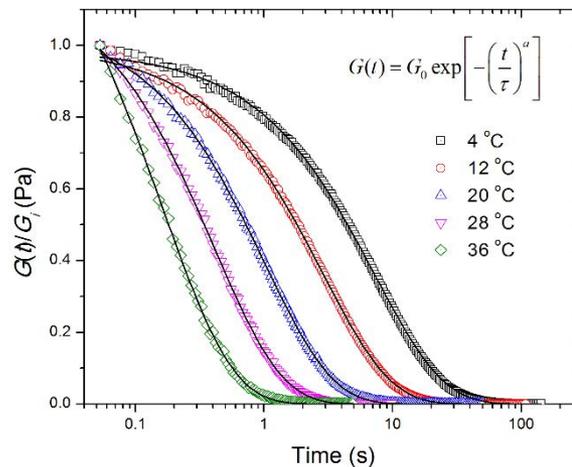
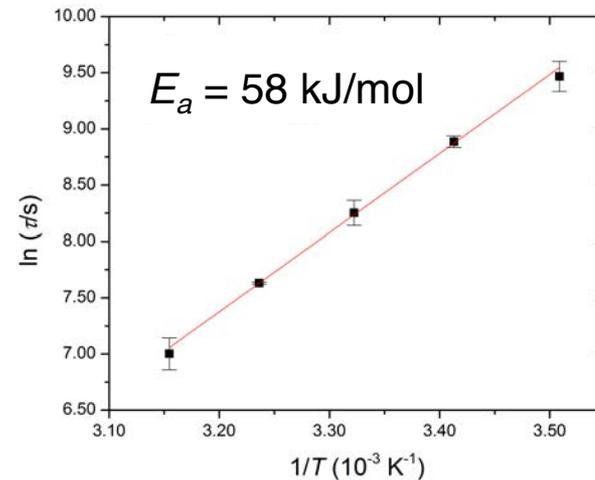
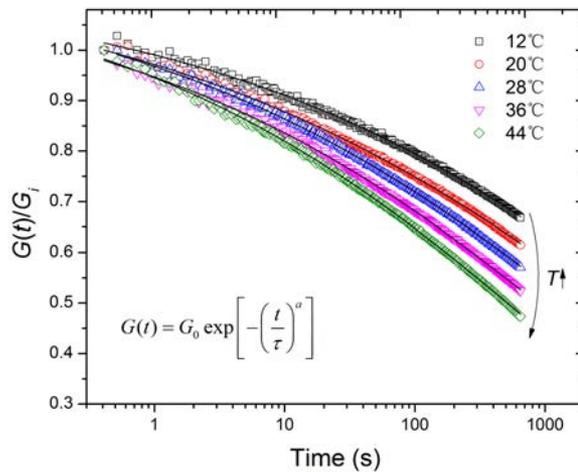


Chemical environment

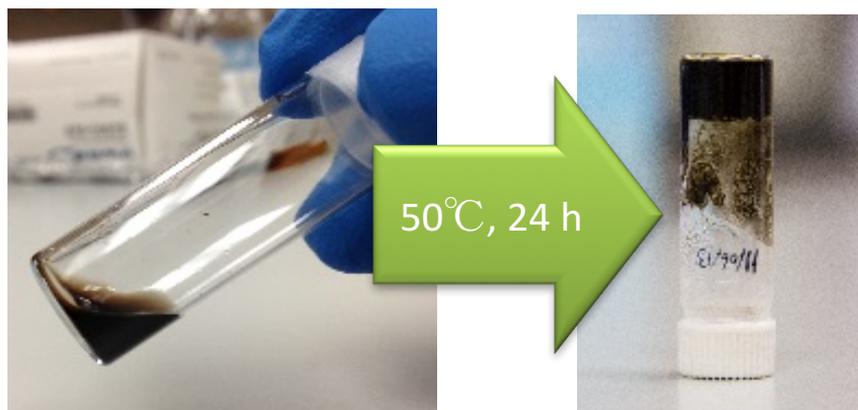
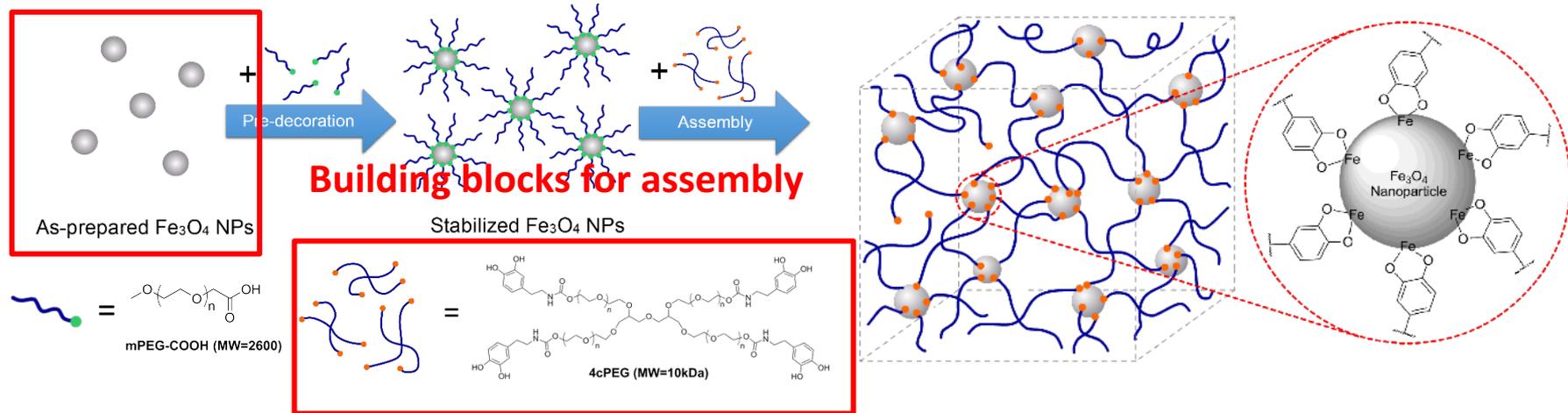


Fe³⁺ gel:

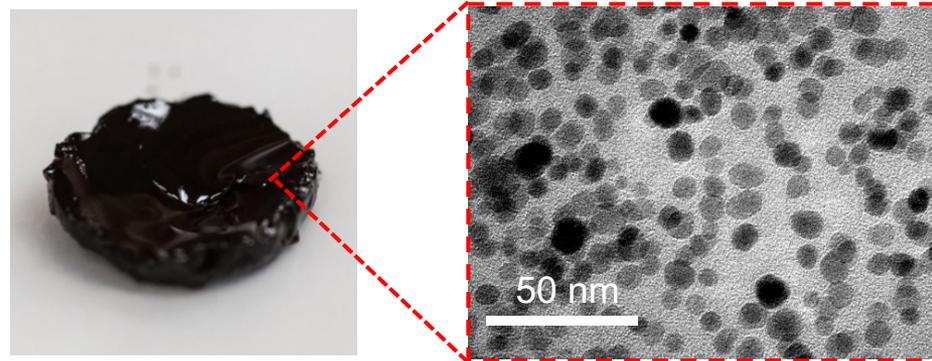
(pH=12)



Metal-coordinate networks: particles



Sol-gel transition



As-prepared NP gel